

16. The dissociation of water is shown below:



- At 25 °C, the auto product of water, K_w , is 1.0×10^{-14} mol² dm⁻⁶.
At 65 °C, the auto product of water, K_w , is 1.0×10^{-13} mol² dm⁻⁶.

17. Explain whether the dissociation of water is an endothermic or exothermic process.
Endothermic.

18. Predict, using a calculation, whether a solution of 1 dm³ of 0.1 M is neutral, acidic or alkaline. (3)

Acidic

19 Explain by calculating pH.

20. K_w , pK_w and pH are significant notions.

Calculate pK_w at 65 °C.

Give your answer to two decimal places.

$$-\log(1.0 \times 10^{-13}) = 13.00$$

$$pK_w = 13.00 \quad (1)$$

21. Describe, at least two ways, how to obtain pure water from sea water without using any chemicals at all.

Calculate the pH of the resulting solution of water at 65 °C.

Give your answer to two decimal places.

$$pH = 7.00$$

pH = _____ (1)
(Total: 10)

Ocr Chemistry A2 Unofficial Mark Scheme 2014

AN Whitehead



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Ocr Chemistry A2 Unofficial Mark Scheme 2014**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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