

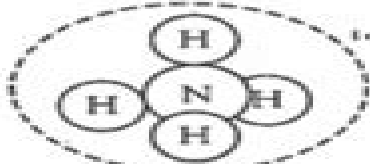
Polyatomic Ions

Can a group of atoms have a charge?

Why?

Do you know you eat a lot of "-ates"? Next time you look at a food label, read the ingredients and you will likely find a number of ingredients that end with "-ate," such as sodium phosphate or calcium carbonate. Did you ever wonder what the chemical formulas of these ingredients look like? In this activity we will explore polyatomic ions, which are groups of atoms that carry a charge. These ions are found in our food ingredients, natural waterways, and many other chemical compounds you encounter every day.

Model 1 – Types of Ions

Monatomic Ions	Nitride 	Sulfide 	Chloride 
Polyatomic Ions	Nitrate 	Sulfate 	Ammonium 
	Nitrite 	Sulfite 	Hydroxide 

1. Use Model 1 to complete the table below.

Name of Ion	Nitride	Nitrate	Sulfate	Sulfite	Ammonium
Charge on Ion	-3	-1	-2	-2	+1
Type and Number of Atoms	1 nitrogen	1 nitrogen 3 oxygen	1 sulfur 4 oxygen	1 sulfur 3 oxygen	1 nitrogen 4 hydrogen
Chemical Formula	N^{3-}	NO_3^{1-}	SO_4^{2-}	SO_3^{2-}	NH_4^{1+}

Polyatomic Ions Pogil Page 3

Melvin Robin



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Investigations Into the Origins of Polyatomic Ions in Inductively Coupled Plasma-mass Spectrometry, 2010 An inductively coupled plasma mass spectrometer ICP MS is an elemental analytical instrument capable of determining nearly all elements in the periodic table at limits of detection in the parts per quadrillion and with a linear analytical range over 8 10 orders of magnitude Three concentric quartz tubes make up the plasma torch Argon gas is spiraled through the outer tube and generates the plasma powered by a looped load coil operating at 27.1 or 40.6 MHz The argon flow of the middle channel is used to keep the plasma above the innermost tube through which solid or aqueous sample is carried in a third argon stream A sample is progressively desolvated atomized and ionized The torch is operated at atmospheric pressure To reach the reduced pressures of mass spectrometers ions are extracted through a series of two approximately one millimeter wide circular apertures set in water cooled metal cones The space between the cones is evacuated to approximately one torr The space behind the second cone is pumped down to or near to the pressure needed for the mass spectrometer MS The first cone called the sampler is placed directly in the plasma plume and its position is adjusted to the point where atomic ions are most abundant The hot plasma gas expands through the sampler orifice and in this expansion is placed the second cone called the skimmer After the skimmer traditional MS designs are employed i.e. quadrupoles magnetic sectors time of flight ICP MS is the leading trace element analysis technique One of its weaknesses are polyatomic ions This dissertation has added to the fundamental understanding of some of these polyatomic ions their origins and behavior Although mainly continuing the work of others certain novel approaches have been introduced here Chapter 2 includes the first reported efforts to include high temperature corrections to the partition functions of the polyatomic ions in ICP MS This and other objections to preceding papers in this area were addressed Errors in the measured T_{gas} values were found for given errors in the experimental and spectroscopic values The ionization energy of the neutral polyatomic ion was included in calculations to prove the validity of ignoring more complicated equilibria Work was begun on the question of agreement between kinetics of the plasma and interface and the increase and depletion seen in certain polyatomic ions This dissertation was also the first to report day to day ranges for T_{gas} values and to use a statistical test to compare different operating conditions This will help guide comparisons of previous and future work Chapter 4 was the first attempt to include the excited electronic state 2 in the partition function of ArO as well as the first to address the different dissociation products of the ground and first electronic levels of ArO Chapter 5 reports an interesting source of memory in ICP MS that could affect mathematical corrections for polyatomic ions For future work on these topics I suggest the following experiments and investigations Clearly not an extensive list they are instead the first topics curiosity brings to mind 1 Measurement of T_{gas} values when using the flow injection technique of Appendix B It was believed that there was a fundamental difference in the plasma when the auto sampler was used versus a continuous injection Is this reflected in T_{gas} values 2 The work of Chapter 3 can be

expanded and supplemented with more trials new cone materials i.e. copper stainless steel and more cone geometries Some of this equipment is already present in the laboratory others could be purchased or made 3 Tgas values from Chapter 3 could be correlated with instrument pressures during the experiment Pressures after the skimmer cone were recorded for many days but have yet to be collated with the measured Tgas values 4 Polyatomic Ions for Surface Analysis and Modification Erick Ryan Fuoco, 2003 Investigations Into the Origins of Polyatomic Ions in Inductively Coupled Plasma-mass Spectrometry Sally M. McIntyre, 2010 **The Particulate Nature of Polyatomic Ions**, 2007 *Fragmentation of Diatomic and Polyatomic Ions in the Gas Phase* Christopher John Proctor, 1981 **Fragmentation of Diatomic and Polyatomic Ions in the Gas Phase** Christopher John Proctor, 1981 Higher Excited States of Polyatomic Molecules V2 Melvin Robin, 2012-12-02 Higher Excited States of Polyatomic Molecules Volume II focuses on a higher level of activity in vacuum ultraviolet spectroscopy This book explores the Rydberg states in atoms and molecules Comprised of five chapters this volume starts with an overview of the two center unsaturated molecules that usually display sharp Rydberg transitions originating with the pi electrons This book then discusses the unsaturated double bond that adds another dimension to the spectrum Other chapters explore the optical spectrum of the amide group which is the basic chromophoric unit in polypeptides This text further discusses the all electron calculations of the electronic structure of the amide group that is performed in Gaussian orbital basis sets This book considers as well the prominent characteristic of Rydberg excitations in benzene The final chapter deals with the biological molecules that are polyfunctional in general Analytical chemists photochemists molecular spectroscopists and researchers will find this book extremely useful **Unimolecular and Collision-induced Dissociation Study of Polyatomic Ions at High Collision Energy** Xuedong Zhou, 2001 **Higher Excited States of Polyatomic Molecules** Melvin Robin, 1974-05-28 Higher Excited States of Polyatomic Molecules Volume I focuses on the spectra in the vacuum ultraviolet region between 50 000 and 100 000 cm⁻¹ This book explores the higher excitations in molecules beyond 50 000 cm⁻¹ Organized into three chapters this volume starts with an overview of the excited state properties of a molecule and the excited state ionization potential This book then proceeds with a discussion of the original classification of the properties as well as the types of excitations observed in the vacuum ultraviolet Other chapters discuss photoelectron spectroscopy which is an independent self sustaining branch of molecular spectroscopy This text examines as well the distinction between valence shell and Rydberg excitations The final chapter deals with several topics including the saturated molecules that are classified as having all valence electrons the alkene absorption spectra and the spectroscopic data on boron compounds Analytical chemists photochemists molecular spectroscopists and researchers will find this book extremely useful **Structure of Free Polyatomic Molecules** K. Kuchitsu, Kozo Kuchitsu, 1998-09-10 This volume Structure of Free Polyatomic Molecules Basic Data contains frequently used data from the corresponding larger Landolt B r nstein handbooks in a low price book for the individual scientists working in the laboratory Directories link to the

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