

Characteristics of Gases

- Unlike liquids and solids, they
 - Expand to fill their containers. (volume of gas = volume of container)
 - Are highly compressible. ($P \propto V$)
 - Have extremely low densities. (much empty space)
 - Form homogeneous mixtures with each other. (diffusion)
- Gases can have very different chemical properties, but their physical properties are very similar.

Section Characteristics Of Gases

Michael Brown



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