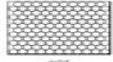
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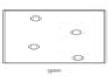
Exercise 1.1

1 A solid has a fixed mass and volume / shape. A liquid has a fixed mass but its shape changes to that of the container in which it is placed. A gas has no fixed shape or volume. A gas completely fills the container that it is in.

2



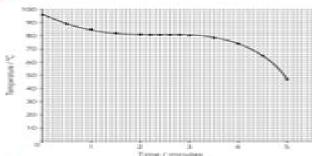




- 3 a f
 - b 6
 - < A.
 - d D
- 4 A freezing (solidification)
 - B mediting
 - C condensation
 - D evaporation or boiling
- 5 a radon
 - b radon and nitrogen
 - nitrogen
 - d cobalt
 - The sample of ethanoic acid is impure.
 The presence of impurities raises the boiling point of a substance.

Exercise 1.2





- 7 The student would need to use an oil bath (in place of the water bath) so that the higher temperature could be reached.
- 8 The substance is freezing (solidifying) / turning from liquid to solid.
- The temperature stays constant because energy is being released as the substance solidifies / the molecules are giving out heat as they stop moving from place to place and become organised in a structured lattice arrangement / in the solid the molecules can only vibrate about fixed points / the heat released by the formation of new interactions (forces) between the particles keeps the temperature constant until all the substance is solid.

Section 8 1 Chemistry Workbook Answers

AN Whitehead

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Table of Contents Section 8 1 Chemistry Workbook Answers

- 1. Understanding the eBook Section 8 1 Chemistry Workbook Answers
 - The Rise of Digital Reading Section 8 1 Chemistry Workbook Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Section 8 1 Chemistry Workbook Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - o Features to Look for in an Section 8 1 Chemistry Workbook Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Section 8 1 Chemistry Workbook Answers
 - Personalized Recommendations
 - Section 8 1 Chemistry Workbook Answers User Reviews and Ratings
 - Section 8 1 Chemistry Workbook Answers and Bestseller Lists

- 5. Accessing Section 8 1 Chemistry Workbook Answers Free and Paid eBooks
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 - Interactive Elements Section 8 1 Chemistry Workbook Answers
- 8. Staying Engaged with Section 8 1 Chemistry Workbook Answers
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 8 1 Chemistry Workbook Answers
- 9. Balancing eBooks and Physical Books Section 8 1 Chemistry Workbook Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 8 1 Chemistry Workbook Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 8 1 Chemistry Workbook Answers
 - Setting Reading Goals Section 8 1 Chemistry Workbook Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 8 1 Chemistry Workbook Answers
 - Fact-Checking eBook Content of Section 8 1 Chemistry Workbook Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

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