Chapter 25 Nuclear Chemistry Practice Test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

	1.	What	does	the 4	in	;He	represent?
--	----	------	------	-------	----	-----	------------

- a. the mass number
- b. the atomic number
- What does the 101 in ²⁵⁶₁₀₁Md represent?
 - a. the mass number
 - b. the atomic number

- c. the nuclide number
- d. the number of neutrons

c. the number of protons

d. the number of neutrons

- b. 239 U

a. nuclear fusion.

c. radioactive decay.

e. electromagnetic waves.

b. nuclear radiation.

d. nuclear fission.

neutrons.

6. Which of the following forms of radiation has the greatest penetrating power?

 a. alpha particles. b. beta particles.

c. gamma rays d. positrons

- Alpha particles are
 - a. electrons.
 - b. helium nuclei.
- 8. Beta particles are
 - a. electrons.
 - b. helium nuclei.

- electromagnetic waves. d. neutrons.
- 9. Which of the illustrations below represents a fission reaction?

Section 4 Review Nuclear Chemistry Answers

Thomas Evangelist

Section 4 Review Nuclear Chemistry Answers:

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