

- Key.
8. How many moles of copper atoms are needed to supply the same number of atoms as does 235 grams of iron?
- $$235 \text{ g} \left(\frac{1 \text{ mol}}{55.8 \text{ g}} \right) = 4.21 \text{ moles of Fe atoms}$$
- 4.21 mols of Cu atoms would have the same number of atoms.
9. How many grams of carbon are needed to supply twice as many atoms as does 27 grams of beryllium?
- $$27 \text{ g} \left(\frac{1 \text{ mol}}{9.0 \text{ g}} \right) = 3.0 \text{ mols Be atoms}$$
- twice atoms $2(3.0 \text{ mols}) = 6.0 \text{ mols}$
- $$6.0 \text{ mols C} \left(\frac{12.0 \text{ g}}{1 \text{ mol}} \right) = 72 \text{ g of carbon.}$$
10. 62.0 grams of rust is composed of 43.4 grams of iron. Determine the empirical formula for rust.
- $$43.4 \text{ g Fe} \rightarrow 43.4 \text{ g} \div 55.8 \text{ g/mol} = 0.78 \text{ mol Fe}$$
- $$62.0 - 43.4 \text{ g O} \rightarrow 18.6 \text{ g} \div 16.0 \text{ g/mol} = 1.16 \text{ mol O}$$
- mol ratio Fe : O
2 : 3
- Fe_2O_3
11. A solution of NaCl contains 35 g of solute dissolved in 100g of water. What is the molarity of the solution?
- $$100 \text{ g H}_2\text{O} \rightarrow 100 \text{ mL H}_2\text{O} \rightarrow 0.100 \text{ L}$$
- $$35 \text{ g} \div 58.5 \text{ g/mol} = 0.60 \text{ mol}$$
- $$M = \frac{n}{V} = \frac{0.60 \text{ mol}}{0.100 \text{ L}} = 6.0 \text{ M}$$
12. The following three containers have identical capacities. Each contains Avogadro's number of molecules of a gaseous element. Fill in the blanks with the missing information.



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J Rink



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Children's Books in Print, 2007, 2006 **A-level Chemistry** E. N. Ramsden, 2000 Each topic is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study **CO2-switchable Materials** Philip G Jessop, Michael F Cunningham, 2020-09-11 CO2 responsive materials are a relatively recent innovation In general stimuli responsive materials exhibit reversible changes in their physical or chemical properties in response to external triggers such as temperature pH light or voltage However there are often limitations in applying these triggers including economic and environmental costs and product contamination The realization that CO2 can be used as an effective trigger for stimulating changes in material properties has prompted a surge in interest in this area within the past few years with numerous new studies currently underway in several countries CO2 is an ideal trigger for switchable or stimuli responsive materials because it is benign inexpensive green abundant and does not accumulate in the system Many different CO2 responsive materials including polymers latexes solvents solutes gels surfactants and catalysts have been prepared Summarizing recent progress in the preparation self assembly and functional applications of CO2 responsive materials this book explores the physical chemistry of CO2 switching including constraints on structural design and process conditions together with applications With emphasis on the environmental health and safety advantages and disadvantages compared to conventional materials it is ideal for researchers and industrialists working in green chemistry chemical engineering and polymer chemistry Chemical Problem-solving by Dimensional Analysis Arnold B. Loebel, 1978 A Comprehensive Treatise on Inorganic and Theoretical Chemistry Joseph William Mellor, 1922 **Supplement to Mellor's Comprehensive Treatise on Inorganic and Theoretical Chemistry** Joseph William Mellor, 1922 **Tietz Textbook of Clinical Chemistry and Molecular Diagnostics - E-Book** Nader Rifai, 2017-01-16 The Tietz Textbook of Clinical Chemistry and Molecular Diagnostics 6th Edition provides the most current and authoritative guidance on selecting performing and evaluating the results of new and established laboratory tests This classic clinical chemistry reference offers encyclopedic coverage detailing everything you need to know including analytical criteria for the medical usefulness of laboratory tests variables that affect tests and results laboratory medicine applications of statistical methods and most importantly clinical utility and interpretation of laboratory tests It is THE definitive reference in clinical chemistry and molecular diagnostics now fully searchable and with quarterly content updates podcasts clinical cases animations and extended content online through Expert Consult Analytical criteria focus on the medical usefulness of laboratory procedures Reference ranges show new

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