#### + WS 6.2 - Nuclear Reactions

Complete each of the following nuclear reactions by determining the missing particle, then name that particle ("alpha particle" or "uranium-233", etc...) #1 is an example...

1. 14C -> 14N + 9e beta particle	symbol ⁴He+≥ or α	name eloha particle
	21.10.2 Of C	alpha particle (a helium nucleus)
2. + 9e> +0Ar	ρe or β	beta particle jan electroni
3. 252C1> + 196Mo + 41n	0er +1	positron (a positive electron?)
	in.	neutron
4. 9Be +> 12C + in	*p or 1H+1	proton (hydrogen-1 nucleus)
5. gLi + 12C> 12F +	2 <b>1-4</b> +1	deutron (hydrogen-2 nucleus)

- 7. 3He + 3He ---> 7Be +
- 8. 235U + 1n --> 144Cs + 90Rb +
- 9. --> 4He + 229Th \_\_\_\_\_
- 10. 39Si + -> 30P \_\_\_\_\_
- 11. 19K + 5n ---> 11Ar +
- 12. 24Mg + -> 22Na + a
- 13. 200Th + β --> 200Po + 2 2H+1 +
- 14. 246Pu + 1n --> 96Mo + + 51n (find the storate #at)
- 15. Write the reaction for a Rn-224 atom undergoing 1 beta decay: What is the daughter?
- 16. Write the reaction for a Po-218 atom undergoing 1 alpha decay: What is the daughter?
- 17. What particle is produced when a U-234 atom undergoes a series of two alpha decays and three beta decays?
- 18. Pb-210 undergoes a series of alpha and beta decays to end up eventually as Ir-198. How many alpha particles and how many beta particles were emitted in all? Hint: do alpha's first, then beta's.  $\alpha = \beta = \beta$

# **Section Radioactivity And Nuclear Reaction Worksheet**

**Peipei Pang** 

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