

7.1

IONS

Section Review

Part A: Completion

Use this completion section to write a four-page summary of the concepts and terms that are introduced in this chapter.

Write under **name**, **number**, **group/period**, **VE**, **electron configuration**, **the number of protons**, **name**, **charge**.

1. Elements within the same group of the periodic table behave similarly because they have the same number of valence electrons.
2. The group number of an element's symbol indicates how many valence electrons that element has.
3. Elements that share valence electrons are called electrons that structure.
4. The most stable described (bonding) configuration exists and is called electron configuration or electron.
5. The number of valence electrons (outermost electrons) of an element is valence configuration changed from valence.
6. The value of Group 1 elements always has a charge of +1.
7. The value of Group 2 elements has a charge of 2.

Part B: True-False

Write True or False underneath the charge that all elements have (VE) or value (VE).

- True 1. The electron value (outermost electrons) of an element is valence.
- True 2. The chemical properties of an element are largely determined by the number of valence electrons the element has.
- True 3. An element's charge is the difference between the number of protons and the number of valence electrons in an atom of that element.
- True 4. The most stable configuration exists for an element when it has the same number of valence electrons as the noble gas nearest it.
- True 5. The number of valence electrons of an element is the same as the number of protons in the nucleus of the element.
- True 6. The number of valence electrons of an element is the same as the number of protons in the nucleus of the element.

Section 7 1 Ions Section Review Answers

Puja Mehta



Section 7 1 Ions Section Review Answers:

Unveiling the Energy of Verbal Art: An Mental Sojourn through **Section 7 1 Ions Section Review Answers**

In a world inundated with monitors and the cacophony of quick communication, the profound energy and psychological resonance of verbal beauty often disappear into obscurity, eclipsed by the continuous barrage of noise and distractions. Yet, nestled within the musical pages of **Section 7 1 Ions Section Review Answers**, a interesting work of fictional elegance that impulses with organic emotions, lies an memorable trip waiting to be embarked upon. Penned with a virtuoso wordsmith, this exciting opus guides readers on an emotional odyssey, lightly exposing the latent possible and profound impact embedded within the intricate web of language. Within the heart-wrenching expanse of the evocative analysis, we will embark upon an introspective exploration of the book is main themes, dissect its fascinating publishing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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