

Chapter 4 Atomic Structure

Section 4.3 Modern Atomic Theory**(pages 113–118)***This section focuses on the arrangement and behavior of electrons in atoms.***Reading Strategy (page 113)**

Sequencing After you read, complete the description in the flow chart below of how the gain or loss of energy affects electrons in atoms. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Bohr's Model of the Atom (pages 113–116)**

1. Circle the letter of the sentence that tells how Bohr's model of the atom differed from Rutherford's model.
 - a. Bohr's model focused on the nucleus.
 - b. Bohr's model focused on the protons.
 - c. Bohr's model focused on the neutrons.
 - d. Bohr's model focused on the electrons.
2. Is the following sentence true or false? In Bohr's model of the atom, electrons have a constant speed and move in fixed orbits around the nucleus. _____
3. What can happen to an electron in an atom when the atom gains or loses energy? _____

4. What evidence do scientists have that electrons can move from one energy level to another? _____

5. Is the following sentence true or false? When electrons release energy, some of the energy may be released as visible light. _____

Electron Cloud Model (page 116)

6. Is the following sentence true or false? Bohr's model was correct in assigning energy levels to electrons. _____
7. When trying to predict the locations and motions of electrons in atoms, scientists must work with _____.
8. What is an electron cloud? _____

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CL Gary



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Table of Contents Section 4 3 Modern Atomic Theory Blank Worksheet

1. Understanding the eBook Section 4 3 Modern Atomic Theory Blank Worksheet
 - The Rise of Digital Reading Section 4 3 Modern Atomic Theory Blank Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 4 3 Modern Atomic Theory Blank Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 4 3 Modern Atomic Theory Blank Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 4 3 Modern Atomic Theory Blank Worksheet

- Personalized Recommendations
- Section 4 3 Modern Atomic Theory Blank Worksheet User Reviews and Ratings
- Section 4 3 Modern Atomic Theory Blank Worksheet and Bestseller Lists
- 5. Accessing Section 4 3 Modern Atomic Theory Blank Worksheet Free and Paid eBooks
 - Section 4 3 Modern Atomic Theory Blank Worksheet Public Domain eBooks
 - Section 4 3 Modern Atomic Theory Blank Worksheet eBook Subscription Services
 - Section 4 3 Modern Atomic Theory Blank Worksheet Budget-Friendly Options
- 6. Navigating Section 4 3 Modern Atomic Theory Blank Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 4 3 Modern Atomic Theory Blank Worksheet Compatibility with Devices
 - Section 4 3 Modern Atomic Theory Blank Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 4 3 Modern Atomic Theory Blank Worksheet
 - Highlighting and Note-Taking Section 4 3 Modern Atomic Theory Blank Worksheet
 - Interactive Elements Section 4 3 Modern Atomic Theory Blank Worksheet
- 8. Staying Engaged with Section 4 3 Modern Atomic Theory Blank Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 4 3 Modern Atomic Theory Blank Worksheet
- 9. Balancing eBooks and Physical Books Section 4 3 Modern Atomic Theory Blank Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 4 3 Modern Atomic Theory Blank Worksheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 4 3 Modern Atomic Theory Blank Worksheet
 - Setting Reading Goals Section 4 3 Modern Atomic Theory Blank Worksheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 4 3 Modern Atomic Theory Blank Worksheet

- Fact-Checking eBook Content of Section 4 3 Modern Atomic Theory Blank Worksheet
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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