

13 RNA and Protein Synthesis

Chapter Test A

Multiple Choice

Write the letter that best answers the question or completes the statement on the line provided.

- _____ 1. Which of the following are found in both DNA and RNA?
- ribose, phosphate groups, and adenine
 - deoxyribose, phosphate groups, and guanine
 - phosphate groups, guanine, and cytosine
 - phosphate groups, guanine, and thymine



Figure 13-1

- _____ 2. Which nucleotide in Figure 13-1 indicates the nucleic acid above is RNA?
- uracil
 - guanine
 - cytosine
 - adenine
- _____ 3. What is produced during transcription?
- RNA molecules
 - DNA molecules
 - RNA polymerase
 - proteins
- _____ 4. During eukaryotic transcription, the molecule that is formed is
- complementary to both strands of DNA.
 - identical to an entire single strand of DNA.
 - double-stranded and inside the nucleus.
 - complementary to part of one strand of DNA.
- _____ 5. There are 64 codons and 20 amino acids. Which of the following is true?
- Several different codons can specify the same amino acid.
 - Each codon specifies a different amino acid.
 - Some amino acids have no link to a codon.
 - Each amino acid is specified by only one codon.
- _____ 6. A promoter is a
- binding site for DNA polymerase.
 - binding site for RNA polymerase.
 - start signal for replication.
 - stop signal for transcription.
- _____ 7. During translation, the type of amino acid that is added to the growing polypeptide depends on the
- codon on the mRNA and the anticodon on the rRNA.
 - anticodon on the mRNA and the anticodon on the tRNA.
 - anticodon on the rRNA and the codon on the mRNA.
 - codon on the mRNA and the anticodon on the tRNA.

Section 3 Dna Rna And Protein Answers

Lingjun Ying



Section 3 Dna Rna And Protein Answers:

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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