

2.1

CHEMICAL REACTIONS

Section Review

Objectives

1. Measure enthalpy using a calorimeter.
2. Identify and predict chemical reactions using phase diagrams.
3. Apply the law of conservation of energy to chemical reactions.

Vocabulary

- | | | |
|-----------------|---------------|----------------------------------|
| 1. endothermic | 2. exothermic | 3. enthalpy |
| 4. fusion curve | 5. product | 6. law of conservation of energy |

Part A Completion

Use the vocabulary words to identify a vocabulary word that matches each given description. Write the word in the space provided.

- | | |
|---|-----------|
| 1. A process that absorbs energy is called a(n) _____. | 1. _____ |
| 2. A process that releases energy is called a(n) _____. | 2. _____ |
| 3. The amount of energy in a system is called _____. | 3. _____ |
| 4. The amount of energy in a system is called _____. | 4. _____ |
| 5. The amount of energy in a system is called _____. | 5. _____ |
| 6. The amount of energy in a system is called _____. | 6. _____ |
| 7. The amount of energy in a system is called _____. | 7. _____ |
| 8. The amount of energy in a system is called _____. | 8. _____ |
| 9. The amount of energy in a system is called _____. | 9. _____ |
| 10. The amount of energy in a system is called _____. | 10. _____ |
| 11. The amount of energy in a system is called _____. | 11. _____ |
| 12. The amount of energy in a system is called _____. | 12. _____ |
| 13. The amount of energy in a system is called _____. | 13. _____ |
| 14. The amount of energy in a system is called _____. | 14. _____ |
| 15. The amount of energy in a system is called _____. | 15. _____ |
| 16. The amount of energy in a system is called _____. | 16. _____ |
| 17. The amount of energy in a system is called _____. | 17. _____ |
| 18. The amount of energy in a system is called _____. | 18. _____ |
| 19. The amount of energy in a system is called _____. | 19. _____ |
| 20. The amount of energy in a system is called _____. | 20. _____ |

Part B True-False

Write true or false in the space provided for each statement. If the statement is false, write the word or words that make it false.

1. The amount of energy in a system is called enthalpy.
2. The amount of energy in a system is called enthalpy.
3. The amount of energy in a system is called enthalpy.

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LP Steffe



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Michael J. Padilla,2002 Sif: Chemistry 5na Tb J. G. R. Briggs,2009 **Holt Chemistry** Ralph Thomas Myers,2004
Applied Mechanics Reviews ,1973 **Fullerenes** Elena Sheka,2011-02-16 At the interface between chemistry biology
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