

Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$				
2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$				
3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$				
4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$				
5) $\text{K} + \text{F} \Rightarrow \text{KF}$				

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B Lingard

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Section 6 Ionic Bonding Worksheet

1. Understanding the eBook Section 6 Ionic Bonding Worksheet
 - The Rise of Digital Reading Section 6 Ionic Bonding Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 6 Ionic Bonding Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 6 Ionic Bonding Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 6 Ionic Bonding Worksheet
 - Personalized Recommendations
 - Section 6 Ionic Bonding Worksheet User Reviews and Ratings
 - Section 6 Ionic Bonding Worksheet and Bestseller Lists
5. Accessing Section 6 Ionic Bonding Worksheet Free and Paid eBooks
 - Section 6 Ionic Bonding Worksheet Public Domain eBooks
 - Section 6 Ionic Bonding Worksheet eBook Subscription Services

- Section 6 Ionic Bonding Worksheet Budget-Friendly Options
- 6. Navigating Section 6 Ionic Bonding Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 6 Ionic Bonding Worksheet Compatibility with Devices
 - Section 6 Ionic Bonding Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 6 Ionic Bonding Worksheet
 - Highlighting and Note-Taking Section 6 Ionic Bonding Worksheet
 - Interactive Elements Section 6 Ionic Bonding Worksheet
- 8. Staying Engaged with Section 6 Ionic Bonding Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 6 Ionic Bonding Worksheet
- 9. Balancing eBooks and Physical Books Section 6 Ionic Bonding Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 6 Ionic Bonding Worksheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 6 Ionic Bonding Worksheet
 - Setting Reading Goals Section 6 Ionic Bonding Worksheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 6 Ionic Bonding Worksheet
 - Fact-Checking eBook Content of Section 6 Ionic Bonding Worksheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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