



REFRIGERATOR

TRUE BUILT IN REFRIGERATOR

BASIC MODEL : RS757LHQESR/**

MODEL NAME : RS27FDBTNSR/AA
RS27FDBTNSR/AC

SERVICE Manual

REFRIGERATOR



RS27FDBTNSR/**

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David Kirk



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web secara umum macam macam dioda ada 7 yaitu dioda zener dioda schottky dioda laser dioda avalanche dioda led light emitting diode photodiode dan juga pn junction diode keseluruhan dari jenis jenis dioda tersebut akan kami kupas dengan penjelasan yang detail melalui ulasan dibawah ini

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web simbol nume descriere diodă dioda permite curgerea curentului numai într o singură direcție de la stânga la dreapta diodă zener permite curgerea curentului într o singură direcție dar de asemenea poate curge în direcția inversă atunci când este peste tensiunea de avarie dioda schottky

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web dioda je elektronički element s dvjema elektrodama kojemu električna vodljivost izrazito ovisi o polaritetu električnog napona između elektroda pa zato propušta električnu struju gotovo samo u jednom smjeru 1 poluvodičke diode se izvode na temelju pn spoja ili na temelju spoja metal poluvodič

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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a al b Ba^{2+} c Br_2 d ca e Ga^{3+} f h 2 g h

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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web an oxidation reduction redox reaction is one in which one or more atoms change oxidation numbers oxidation occurs when an atom s oxidation state becomes more positive indicating that electrons have been lost reduction occurs when an atom s oxidation state becomes less positive indicating that electrons have been gained

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web describe redox reactions in terms of electron transfer recall that oxidation involves loss of electrons and reduction involves gain of electrons represent simple redox reactions using half equations identify oxidising and reducing agents for simple redox reactions understand the reactivity of metals and metal displacement reaction

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 $\text{Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$
oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3\text{Ag}$ oxidation reduction 25 $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 O_2

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