

Science

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(Chapter – 1) (Chemical Reactions and Equations) (Exercises) (Class 10)

Question 1:

Which of the statements about the reaction below are incorrect?



- (a) Lead is getting reduced.
- (b) Carbon dioxide is getting oxidised.
- (c) Carbon is getting oxidised.
- (d) Lead oxide is getting reduced.

(i) (a) and (b)

(ii) (a) and (c)

(iii) (a), (b) and (c)

(iv) all

Answer 1:

(i) (a) and (b)

Question 2:

The above reaction is an example of a



- (a) combination reaction.
- (c) decomposition reaction.

- (b) double displacement reaction.
- (d) displacement reaction.

Answer 2:

(d) displacement reaction.

Question 3:

What happens when dilute hydrochloric acid is added to iron filings? Tick the correct answer.

- (a) Hydrogen gas and iron chloride are produced.
- (c) No reaction takes place.

- (b) Chlorine gas and iron hydroxide are produced.
- (d) Iron salt and water are produced.

Answer 3:

(a) Hydrogen gas and iron chloride are produced.

Question 4:

What is a balanced chemical equation? Why should chemical equations be balanced?

Answer 4:

A reaction which has an equal number of atoms of all the elements on both sides of the chemical equation is called a balanced chemical equation.

The law of conservation of mass states that mass can neither be created nor destroyed. Hence, in a chemical reaction, the total mass of reactants should be equal to the total mass of the products. It means that the total number of atoms of each element should be equal on both sides of a chemical equation. Hence, it is for this reason that chemical equations should be balanced.

Question 5:

Translate the following statements into chemical equations and then balance them.

- (a) Hydrogen gas combines with nitrogen to form ammonia.
- (b) Hydrogen sulphide gas burns in air to give water and sulphur dioxide.
- (c) Barium chloride reacts with aluminium sulphate to give aluminium chloride and a precipitate of barium sulphate.
- (d) Potassium metal reacts with water to give potassium hydroxide and hydrogen gas.

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Dr. K. G. Ojha



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Table of Contents **Ncert 1chemistry Chapter Wise Notes**

1. Understanding the eBook **Ncert 1chemistry Chapter Wise Notes**
 - The Rise of Digital Reading **Ncert 1chemistry Chapter Wise Notes**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Ncert 1chemistry Chapter Wise Notes**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **Ncert 1chemistry Chapter Wise Notes**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Ncert 1chemistry Chapter Wise Notes**
 - Personalized Recommendations
 - **Ncert 1chemistry Chapter Wise Notes** User Reviews and Ratings
 - **Ncert 1chemistry Chapter Wise Notes** and Bestseller Lists

5. Accessing Ncert 1chemistry Chapter Wise Notes Free and Paid eBooks
 - Ncert 1chemistry Chapter Wise Notes Public Domain eBooks
 - Ncert 1chemistry Chapter Wise Notes eBook Subscription Services
 - Ncert 1chemistry Chapter Wise Notes Budget-Friendly Options
6. Navigating Ncert 1chemistry Chapter Wise Notes eBook Formats
 - ePub, PDF, MOBI, and More
 - Ncert 1chemistry Chapter Wise Notes Compatibility with Devices
 - Ncert 1chemistry Chapter Wise Notes Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ncert 1chemistry Chapter Wise Notes
 - Highlighting and Note-Taking Ncert 1chemistry Chapter Wise Notes
 - Interactive Elements Ncert 1chemistry Chapter Wise Notes
8. Staying Engaged with Ncert 1chemistry Chapter Wise Notes
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ncert 1chemistry Chapter Wise Notes
9. Balancing eBooks and Physical Books Ncert 1chemistry Chapter Wise Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ncert 1chemistry Chapter Wise Notes
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ncert 1chemistry Chapter Wise Notes
 - Setting Reading Goals Ncert 1chemistry Chapter Wise Notes
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ncert 1chemistry Chapter Wise Notes
 - Fact-Checking eBook Content of Ncert 1chemistry Chapter Wise Notes
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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