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EXPERIMENTS)

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CHEMISTRY

1.3	pH	of Sici	lutions	11-25
	(AV)	20.6	nd the p41 of the following samples by using p11 paper/universal indicator:	13
		(0)	Dilute hydrochloric acid (HCI)	
		0.0	Diliete sedium kydroxide (NaOH) solution	
		(6)(6)	Dilute ethanoic acid (CH ₂ COOH) solution	
		((iv)	Lemon Juice	
		(v)	Winter	
		$(\mathbf{v}(i))$	Dilute hydrogen carbonate solution.	
	(89)	Street	lying the properties of acids and bases (EICI and NaOII) on the basis of their reaction with	
			and the state of t	
		(40)	Litmus solution (blue)red)	
		(05)	Zinc metal	
		(60)	Solid sollium carbonate.	
2.	Typ	es of	Chemical Reactions	26-42
	60	Zo p	ceform and obscree the action of seater on quick lime and classify the reaction.	27
	640	Top	reform and observe the action of heat on farrous sulphate cryotals and classify the reaction.	34
	0=0		regions and observe the reaction of iron noils kept in capper sulplante solution and el- tion.	mojfy the
	-000	250 m	enform and observe the neaction between sodium sulphate and barison chievide solutions as	of chances
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