

**Organic Chemistry 32-235  
Practice Questions for Exam #2**

Part 1: (Circle only **ONE** choice, circling more than one will be counted as wrong!)  
4 points each

1. The correct IUPAC name for the following compound is:

- (A) (1R, 3R)-1-chloro-3-methylcyclohexane
- (B) (1R, 3S)-1-chloro-3-methylcyclohexane
- (C) (1S, 3S)-1-chloro-3-methylcyclohexane
- (D) (1S, 3R)-1-chloro-3-methylcyclohexane
- (E) None of the above.



The Key here is to determine the configuration R or S for each chiral carbon. The answer here is 1S, 2S.

2. Consider the  $S_N1$  reaction of tert-butyl chloride with iodide ion:

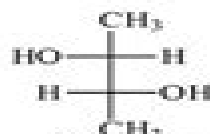
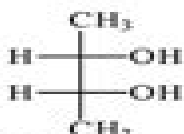


If the concentration of iodide ion is doubled, the rate of forming tert-butyl iodide will:  
(hint: consider mechanism, i.e. how is the product formed?)

- (A) Double.
- (B) Increase 4 times.
- (C) Remain the same.
- (D) Decrease.
- (E) None of the above.

In  $S_N1$  reaction, the rate is independent of the nucleophile involved since the nucleophile is not involved in the rate determining step. Thus the answer is C.

3. The relationship between the following two structures is:



- (A) enantiomers
- (B) diastereomers
- (C) structural isomers
- (D) identical
- (E) none of the above

The key here is to know the definition of all terms and then determine R or S configuration for each chiral carbon. The answer is (B) (by definition)

4. The specific rotation of pure (R)-2-butanol is  $-13.5^\circ$ . What % of a mixture of the two enantiomeric forms is (S)-2-butanol if the specific rotation of this mixture is  $-5.4^\circ$ ?

- (A) 40%
- (B) 30%
- (C) 60%
- (D) 70%
- (E) None of the above

The answer is B since e.e. is 40% favoring R. ( $5.4/13.5=40\%$ )

# Practice Questions Topic 2 Answer Chemistry

**ML Yell**



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## **Table of Contents Practice Questions Topic 2 Answer Chemistry**

1. Understanding the eBook Practice Questions Topic 2 Answer Chemistry
  - The Rise of Digital Reading Practice Questions Topic 2 Answer Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Practice Questions Topic 2 Answer Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Practice Questions Topic 2 Answer Chemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Practice Questions Topic 2 Answer Chemistry
  - Personalized Recommendations
  - Practice Questions Topic 2 Answer Chemistry User Reviews and Ratings
  - Practice Questions Topic 2 Answer Chemistry and Bestseller Lists
5. Accessing Practice Questions Topic 2 Answer Chemistry Free and Paid eBooks
  - Practice Questions Topic 2 Answer Chemistry Public Domain eBooks
  - Practice Questions Topic 2 Answer Chemistry eBook Subscription Services
  - Practice Questions Topic 2 Answer Chemistry Budget-Friendly Options
6. Navigating Practice Questions Topic 2 Answer Chemistry eBook Formats
  - ePub, PDF, MOBI, and More
  - Practice Questions Topic 2 Answer Chemistry Compatibility with Devices
  - Practice Questions Topic 2 Answer Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Practice Questions Topic 2 Answer Chemistry
  - Highlighting and Note-Taking Practice Questions Topic 2 Answer Chemistry
  - Interactive Elements Practice Questions Topic 2 Answer Chemistry
8. Staying Engaged with Practice Questions Topic 2 Answer Chemistry

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Practice Questions Topic 2 Answer Chemistry
- 9. Balancing eBooks and Physical Books Practice Questions Topic 2 Answer Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Practice Questions Topic 2 Answer Chemistry
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Practice Questions Topic 2 Answer Chemistry
  - Setting Reading Goals Practice Questions Topic 2 Answer Chemistry
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Practice Questions Topic 2 Answer Chemistry
  - Fact-Checking eBook Content of Practice Questions Topic 2 Answer Chemistry
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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