Answer to Mole Problems Mole Worksheet KEY Molar Relationships

Name

Directions: Answer the following questions **on a separate sheet of paper.** Set-up all problems using the factor-label method of dimensional analysis and show all your work and units.

What is the mass of 7.50 moles of sulfur dioxide (SO₂)?480g

$$\frac{7.5mol}{1}x\frac{64g}{1mol} = 480g$$

2. How many moles are there in 250.0 grams of sodium phosphate (Na₃PO₄)? 1.52mol

$$\frac{250g}{1}x\frac{1mol}{164g} = 1.52mol$$

How many grams of potassium sulfate (K₂SO₄) are there in 25.3 moles? 4402g

$$\frac{25.3mol}{1} \times \frac{176g}{1mol} = 4402g$$

4. What is the volume of 0.38 moles of any gas at STP? 8.5L

$$\frac{0.38mol}{1}x\frac{22.4L}{1mol} = 8.5L$$

7. What is the mass of 51 liters of oxygen gas? 73g

$$\frac{51L}{1}x\frac{1mol}{22.4L}x\frac{32g}{1mol} = 73g$$

1. Calculate the number of moles in 32.2-L of NH₃

$$\frac{32.3L}{1}x\frac{1mol}{22.4L} = 1.44mol$$

Calculate the number of grams in 3.25-mol of AgNO₃

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Marion F. Gooding

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