

A Periodic Table Logic Problem

Purpose: In this activity you will use a set of clues to identify a set of unknown elements in a periodic table. The elements are the elements with atomic numbers 1-20 and 31-36. A letter of the alphabet is used to represent each unknown element. The letter designation is not related to an element's chemical symbol. Each clue refers to a property of an element or a relationship an element has to other elements in the periodic table. Along with logic and knowledge of properties, you will use the periodic trends to solve the puzzle.

Grade: Neatness and completeness count.

Procedure: Use the following trend clues to place the elements in their proper places on the provided periodic table. Try to identify the elements with direct clues first. For example, B is a direct clue because it identifies only one element.
Group 1 Group 2 Group 13 Group 14 Group 15 Group 16 Group 17 Group 18

Work copy

Group 1	Group 2		Group 13	Group 14	Group 15	Group 16	Group 17	Group 18
–								–
–	–	–	–	–	–	–	–	–
–	–		–	–	–	–	–	–
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A – Has one of the highest electronegativities on the table that is NOT a halogen.

B – Has one electron in a 3p orbital.

C – Has five electrons in the 4th energy level.

D – Forms the smallest 2+ ion.

E – Transition Metals are found in these groups

F – Electron configuration is $1s^2 2s^2 2p^6 3s^2 3p^3$

G – The most electronegative element.

H – An ion of this element with a 2+ charge has 18 electrons.

I – This group is known as the Halogens

J – It's shorthand electron configuration must have element M in brackets

Periodic Table Logic Problem

M Tight



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Table of Contents Periodic Table Logic Problem

1. Understanding the eBook Periodic Table Logic Problem
 - The Rise of Digital Reading Periodic Table Logic Problem
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Logic Problem
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Logic Problem
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Logic Problem
 - Personalized Recommendations
 - Periodic Table Logic Problem User Reviews and Ratings
 - Periodic Table Logic Problem and Bestseller Lists
5. Accessing Periodic Table Logic Problem Free and Paid eBooks
 - Periodic Table Logic Problem Public Domain eBooks
 - Periodic Table Logic Problem eBook Subscription Services
 - Periodic Table Logic Problem Budget-Friendly Options

6. Navigating Periodic Table Logic Problem eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Logic Problem Compatibility with Devices
 - Periodic Table Logic Problem Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Logic Problem
 - Highlighting and Note-Taking Periodic Table Logic Problem
 - Interactive Elements Periodic Table Logic Problem
8. Staying Engaged with Periodic Table Logic Problem
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Logic Problem
9. Balancing eBooks and Physical Books Periodic Table Logic Problem
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Logic Problem
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Logic Problem
 - Setting Reading Goals Periodic Table Logic Problem
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Logic Problem
 - Fact-Checking eBook Content of Periodic Table Logic Problem
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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