

Periodic Trends Worksheet

Using a few minutes before today's activity, complete the following questions. Please post your responses on the board when you are done.

Additional help on the periodic trends can be found on this video: <https://www.youtube.com/watch?v=0D8Wp-2GQp0>

1. What does electronegativity measure and how does it change across the periodic table (increased or decreased)? Explain, reference, substantiate, and justify.
2. Ionization energy increases from left to right across the periodic table. Explain.
3. Ionization energy increases from left to right across the periodic table. Explain your answer.
 - a. Na or Li
 - b. Cl or Br
 - c. Cl or C
4. Ionization energy increases from left to right across the periodic table. Explain your answer.
 - a. R or Ca
 - b. Ca or Fe
 - c. Li or Fe
5. Ionization energy increases from left to right across the periodic table. Explain your answer.
 - a. Cl or C
 - b. Na or Li
 - c. R or Fe
6. Ionization energy increases from left to right across the periodic table. Explain your answer.
 - a. Na or C
 - b. Na or Fe
 - c. R or Fe
7. Ionization energy increases from left to right across the periodic table. Explain your answer.
 - a. R or Fe
 - b. Ca or Fe
 - c. Na or Fe
8. Atomic radius increases from left to right across the periodic table. Explain your answer.
 - a. Na or C
 - b. Na or Fe
 - c. R or Fe
9. Atomic radius increases from left to right across the periodic table. Explain your answer.
 - a. R or Fe
 - b. Ca or Fe
 - c. Na or Fe
10. Which is the most electronegative element on the periodic table? Which is the least electronegative?
 - a.
 - b.

Periodic Trends Questions And Answers

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