

Name: _____

Dates:

CHEMISTRY

Periodic Table of the Elements

Use the Periodic Table of the Elements to answer the questions.

The image shows a standard periodic table of elements. The elements are color-coded by groups:

- Group 1 (Alkali metals): Light blue
- Group 2 (Alkaline earth metals): Orange
- Groups 3-10 (Transition metals): Yellow
- Group 11 (Coinage metals): Light green
- Group 12 (Zn, Cd, Hg): Light blue
- Group 13 (Boron group): Light orange
- Group 14 (Carbon group): Light green
- Group 15 (Nitrogen group): Light blue
- Group 16 (Chalcogens): Light orange
- Group 17 (Halogens): Light green
- Group 18 (Noble gases): Light blue
- Group 19 (Alkali metals): Light blue
- Group 20 (Alkaline earth metals): Orange
- Group 21 (Transition metals): Yellow
- Group 22 (Transition metals): Light green
- Group 23 (Transition metals): Light blue
- Group 24 (Transition metals): Light orange
- Group 25 (Transition metals): Light green
- Group 26 (Transition metals): Light blue
- Group 27 (Transition metals): Light orange
- Group 28 (Transition metals): Light green
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- Group 112 (Transition metals): Light green
- Group 113 (Transition metals): Light blue
- Group 114 (Transition metals): Light orange
- Group 115 (Transition metals): Light green
- Group 116 (Transition metals): Light blue
- Group 117 (Transition metals): Light orange
- Group 118 (Transition metals): Light green
- Group 119 (Transition metals): Light blue
- Group 120 (Transition metals): Light orange

Labels for Boron (B) and its isotopes are present in the top right corner of the image:

- B**: Boron
- 10.811**: Atomic weight
- 5**: Atomic number
- Isotopes**: A list of isotopes with their half-lives and decay modes:
 - ^{10}B (stable)
 - ^{11}B (stable)
 - ^{12}B (12.5 min, β^-)
 - ^{13}B (1.37 min, β^-)
 - ^{14}B (1.25 min, β^-)
 - ^{15}B (0.0117 s, β^-)
 - ^{16}B (0.00012 s, β^-)
 - ^{17}B (0.00003 s, β^-)
 - ^{18}B (0.000001 s, β^-)
 - ^{19}B (0.0000001 s, β^-)
 - ^{20}B (0.00000001 s, β^-)
 - ^{21}B (0.000000001 s, β^-)
 - ^{22}B (0.0000000001 s, β^-)
 - ^{23}B (0.00000000001 s, β^-)
 - ^{24}B (0.000000000001 s, β^-)
 - ^{25}B (0.0000000000001 s, β^-)
 - ^{26}B (0.00000000000001 s, β^-)
 - ^{27}B (0.000000000000001 s, β^-)
 - ^{28}B (0.0000000000000001 s, β^-)
 - ^{29}B (0.00000000000000001 s, β^-)
 - ^{30}B (0.000000000000000001 s, β^-)
 - ^{31}B (0.0000000000000000001 s, β^-)
 - ^{32}B (0.00000000000000000001 s, β^-)
 - ^{33}B (0.000000000000000000001 s, β^-)
 - ^{34}B (0.0000000000000000000001 s, β^-)
 - ^{35}B (0.00000000000000000000001 s, β^-)
 - ^{36}B (0.000000000000000000000001 s, β^-)
 - ^{37}B (0.0000000000000000000000001 s, β^-)
 - ^{38}B (0.00000000000000000000000001 s, β^-)
 - ^{39}B (0.000000000000000000000000001 s, β^-)
 - ^{40}B (0.0000000000000000000000000001 s, β^-)
 - ^{41}B (0.00000000000000000000000000001 s, β^-)
 - ^{42}B (0.000000000000000000000000000001 s, β^-)
 - ^{43}B (0.0000000000000000000000000000001 s, β^-)
 - ^{44}B (0.00000000000000000000000000000001 s, β^-)
 - ^{45}B (0.000000000000000000000000000000001 s, β^-)
 - ^{46}B (0.0000000000000000000000000000000001 s, β^-)
 - ^{47}B (0.00000000000000000000000000000000001 s, β^-)
 - ^{48}B (0.000000000000000000000000000000000001 s, β^-)
 - ^{49}B (0.0000000000000000000000000000000000001 s, β^-)
 - ^{50}B (0.00000000000000000000000000000000000001 s, β^-)
 - ^{51}B (0.000000000000000000000000000000000000001 s, β^-)
 - ^{52}B (0.000

Part 1: Atomic Number. On the line next to the element, write the atomic number of the element. Your answer should be 1, 2, or 3 digits. No letters.

- | | | | | | |
|----|--|-----------|-----|--|--------------|
| 1. | | Hydrogen | 7. | | Cerium |
| 2. | | Oxygen | 8. | | Iridium |
| 3. | | Magnesium | 9. | | Bismuth |
| 4. | | Vanadium | 10. | | Uranium |
| 5. | | Krypton | 11. | | Fermium |
| 6. | | Ruthenium | 12. | | Darmstadtium |

Periodic Table Answers Section 3

Mark Steyvers

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web all room air exhausted directly to outdoors air recirculated by room units design relative humidity design temp c operating room positive 4 20 nr no 20 to 60 20 to 24 emergency department public waiting area negative 2 12 yes nr max 65 21 to 24 aii rooms negative 2 12 yes no max 60 21 to 24 patient room nr 2 4 nr nr max 60

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effect at the time of the ventilation system installation

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web apr 23 2008 1 3 8 the temperature setting on forced air warming devices should be set at maximum and then adjusted to maintain a patient temperature of at least 36 5 c 2008 1 3 9 all irrigation fluids used intraoperatively should be warmed in a thermostatically controlled cabinet to a temperature of 38 c to 40 c 2008

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and ventilation standards some areas are considered critical such as operating rooms isolation rooms and *air humidification in hospitals and healthcare structures with the* - Jun 22 2022

web to control both air temperature and humidity accurate humidity control also helps protect electronic equipment against electrostatic discharges ensuring a safer place for personnel and visitors and at the same time safeguarding patient health standards for operating rooms european directive 2002 91 ec and the en iso

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web mar 2 2020 each space and any special clinical needs for temperature humidity and pressure control the program shall include space names and paragraph numbering references from the applicable version of the fgi guidelines for design and construction of hospitals and outpatient facilities for each space noted within the program this

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