



Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H₂O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

Percent Yield Answers

EW Minium



Percent Yield Answers:

Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation *To Review H.R. 4351, as Amended and Reported by the Committee on Interior and Insular Affairs* United States. Congress. House. Committee on Armed Services. Subcommittee on Military Installations and Facilities, 1987

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An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

Green Chemistry and Engineering Anne E. Marteel-Parrish, Martin A. Abraham, 2013-11-11 Promotes a green approach to chemistry and chemical engineering for a sustainable planet With this text as their guide students will gain a new outlook on chemistry and engineering The text fully covers introductory concepts in general organic inorganic and analytical chemistry as well as biochemistry At the same time it integrates such concepts as greenhouse gas potential alternative and renewable energy solvent selection and recovery and ecotoxicity As a result students learn how to design chemical products and processes that are sustainable and environmentally friendly Green Chemistry and Engineering presents the green approach as an essential tool for tackling problems in chemistry A novel feature of the text is its integration of introductory engineering concepts making it easier for students to move from fundamental science to applications Throughout this text the authors integrate

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A Teacher's Guide to Conversational AI David A. Joyner, 2024-05-27 A Teacher's Guide to Conversational AI explores the practical role that language based artificial intelligence tools play in classroom teaching learning experiences and student assessment Today's educators are well aware that conversational and generative AI chatbots intelligent tutoring systems large language models and more represent a complex new factor in teaching and learning This introductory primer offers comprehensive novice friendly guidance into the challenges and opportunities of incorporating AI into K 12 schools and college classes in ways that are appropriate nourishing to students and outcomes driven Opening with an informative overview of the foundational properties key terminology and ethical considerations of these tools the book offers a coherent and realistic vision of classrooms that are enhanced rather than stymied by AI systems This includes strategies for designing assessments that are conducive to students beneficial use of AI while mitigating overreliance or dishonesty using AI to generate lesson examples for student critique or custom content that reinforces course principles leveraging chatbots as a co instructor or a tutor a guide during student driven learning a virtual debate or brainstorming partner and a design project and creating course content lesson plans and activities expanded language and accessibility options and beyond Through the depth of understanding and applied approach provided in these chapters teachers and leaders in training and in service alongside private tutors college instructors and other educators will be better prepared to future proof their efforts to serve new generations of learners

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Foundations of College Chemistry Morris Hein,Susan Arena,Cary Willard,2016-08-02

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Resource Recovery Implementation United States. Congress. House. Committee on Interstate and Foreign Commerce,1977

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