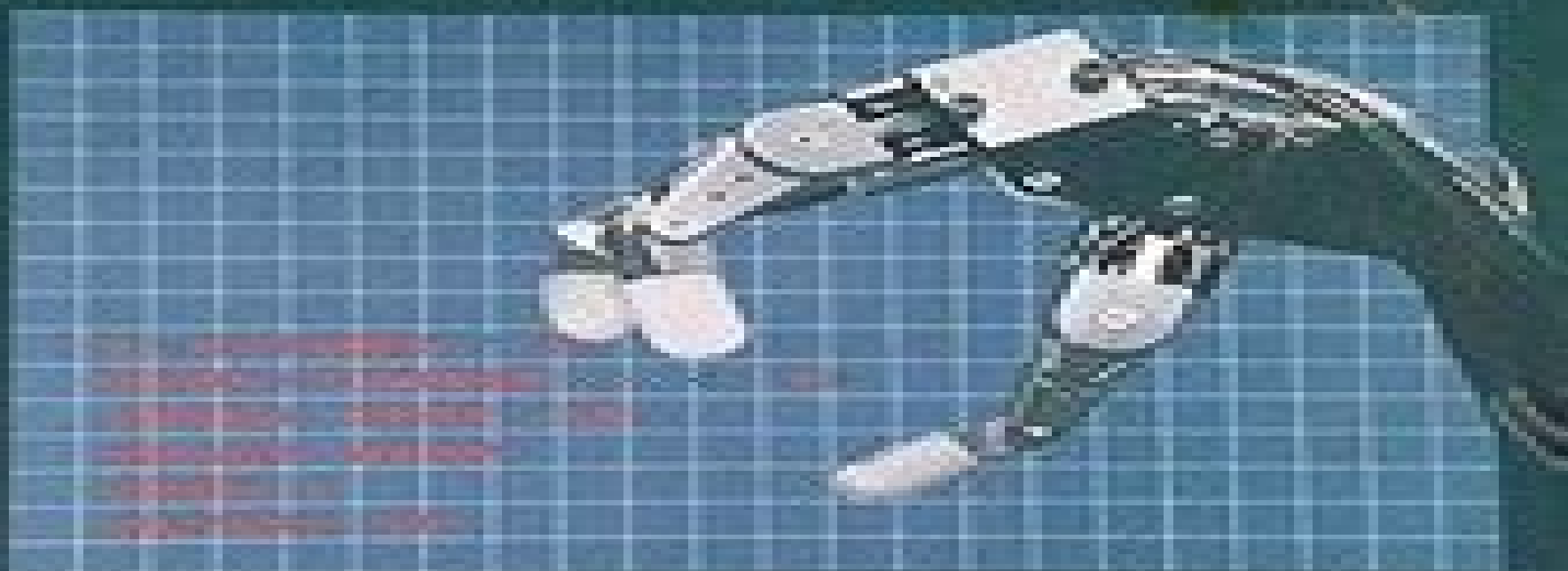


# ROBOTICA INDUSTRIAL




## TECNOLOGIA, PROGRAMACION Y APLICACIONES

Mikael E. Groover, Michael Mass,  
Roger M. Nagel y Nicholas G. Odrey



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web acs practice test 1 acids bases 1 all are examples of lewis acid base reactions except a b c d  $\text{Cu}^{2+} + 4\text{NH}_3 \rightleftharpoons \text{Cu}(\text{NH}_3)_4^{2+}$  e  $\text{HCl} + \text{NH}_3 \rightleftharpoons \text{NH}_4\text{Cl}$  f  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  g  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  h  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  i  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  j  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  k  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  l  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  m  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  n  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  o  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  p  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  q  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  r  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  s  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  t  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  u  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  v  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  w  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  x  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  y  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$  z  $\text{H}_2\text{O} + \text{Na} \rightleftharpoons \text{NaOH} + \text{H}_2$

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