

Colligative Properties Worksheet

FYI: K_f of water is 1.86°C/m ; K_b of water is 0.52°C/m

- 1) Explain why the addition of a solute decreases the melting point of a liquid.
- 2) What is the melting point of a solution in which 3.5 grams of sodium chloride is added to 230 mL of water?
- 3) What is the boiling point of a solution in which 22.5 grams of sucrose ($\text{C}_{12}\text{H}_{22}\text{O}_{11}$) is dissolved in 550 mL of water?
- 4) Rank the following solutions from lowest to highest boiling point, a 2.5 m solution of sodium chloride, a 3.5 m solution of magnesium chloride, or a 4.5 m solution of sulfur dioxide?
- 5) If an aqueous solution has a boiling point of 102.5° , what's the effective molality of the solute?

Practice Test On Colligative Properties

Albert A Gayle



Practice Test On Colligative Properties:

Reviewing **Practice Test On Colligative Properties**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Practice Test On Colligative Properties**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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