

5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

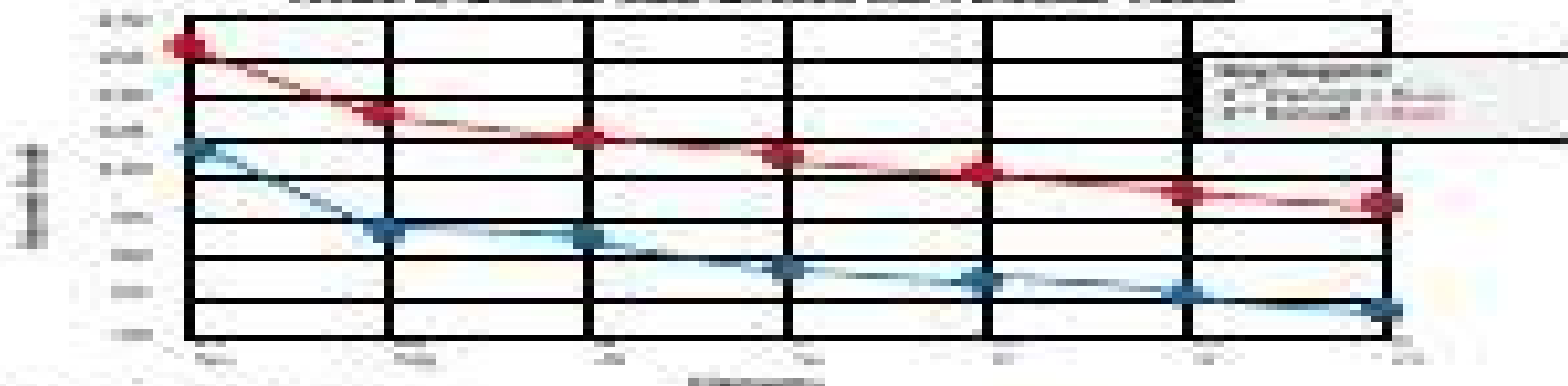
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property. Sketch it's data points (graphical)
2. Is a slight downward trend in this graph? Period? If not, look at the other graphs.
3. Label the box as 'not suitable' (highlighted)
4. Answer all questions...

	1	2	3	4	5	6	7	8
1	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
2	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
3	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
4	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
5	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
6	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
7	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
8	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
9	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity

Now label the trends in the following table:

Trend in Atomic Size across the Periodic Table



Electronegativity Questions

1. Following the trend in atomic size across the periodic table, decrease
2. This means two more than just mean decrease (periods) differences. They are all decreasing (periods) differences across the periodic table.

Electronegativity Questions

1. Electronegativity trend in the periodic table is decreasing (periods) differences. It is not all decreasing (periods) differences across the periodic table.
2. This means two more than just mean decrease (periods) differences. They are all decreasing (periods) differences across the periodic table.

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SA Dillow



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