Periodic Trends

Directions: Choose the letter that corresponds to the correct answer.

- Which of the following can be considered as Dobereiner's triad?
 - a. He: 4.0. Lt: 6.9. Na: 23.0
 - b. N: 14.0, P: 40.0, At 217.0
 - Ne: 20.2, Ar. 39.9, Kr. 83.8
 - d. Ca: 40.1, Sr. 87.6, Ba: 137.3
- 2. Which of the following elements does not belong to the group?
 - a. N
 - b S
 - c. Po
 - d. Ter
- 3. Which of the following is/are TRUE about boron and oxygen?
 - a. both are metalloids
 - b. boron is smaller than oxygen.
 - boron is more electronegative than oxygen
 - d. higher energy is required to eject an electron from the valence shell of boron.
 - e. all of the above
 - f. none of the above

For 4 and 5, consider Na, Mg Al, and Si.

- The element that ejects one electron from its valence shell most readily is
 - a Na
 - b. Ma
 - C. All
 - d. Si
- 5. Which element has the most negative electron affinity?
 - e. Na
 - f. Mg



Periodic Trends Practice Questions

TD Snyder

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