

Name: _____

Period: _____

Periodic Table WebQuest

Instructions: In each of the websites listed, find the information asked for.



Part I: "Get Organized Periodically"

<http://www.chemicalsciences.com/lesson/periodic.html>

1. Why are the elements placed in specific places on the Periodic Table?
2. Periods are _____ that run from left to right.
3. Elements in the same period have the same _____.
4. Every element in the first period has _____ shell for its _____. Every element in the second period has _____ for its _____. See the pattern?
5. Groups are _____ that run from top to bottom.
6. The elements of a group have the same number of _____ in their _____ shell.
7. Every element in group one has _____ electron in its outer shell. Every element in group two has _____ electrons in its outer shell.
8. Hydrogen is special because it can act like two groups, _____ and _____.
9. Hydrogen sometimes is _____ an electron and sometimes it has an _____ electron.
10. Although helium has only _____ electrons in its outer shell, it is grouped with elements that have _____.
11. The green elements on this table are called _____ elements. They each have two electrons in their outer shell.

Part II: "Family Fun" Go to <http://chemicalsciences.com/>

12. Click on Alkali Metals (left bar) and answer the following questions.
 - a. What is the group number? _____
 - b. Are these metals reactive? _____
 - c. Do these metals occur freely in nature? _____
 - d. How many electrons are in their outer shell? _____

Periodic Table Webquest Answer Key

C Cleary



Periodic Table Webquest Answer Key:

Unveiling the Energy of Verbal Art: An Mental Sojourn through **Periodic Table Webquest Answer Key**

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