

s- block		2.7 Table : Modern Periodic Table																p- block					
		Atomic No. Symbol Name Atomic mass																					
1 H Hydrogen 1.008	2 He Helium 4.003																	13 B Boron 10.81	14 C Carbon 12.011	15 N Nitrogen 14.007	16 O Oxygen 15.999	17 F Fluorine 18.998	18 Ne Neon 20.180
3 Li Lithium 6.941	4 Be Beryllium 9.012	d- block																31 Al Aluminum 26.982	32 Si Silicon 28.086	33 P Phosphorus 30.974	34 S Sulfur 32.06	35 Cl Chlorine 35.45	36 Ar Argon 39.948
11 Na Sodium 22.990	12 Mg Magnesium 24.305	21 Sc Scandium 44.956	22 Ti Titanium 47.867	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.798						
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.906	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.757	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29						
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71 f- block	72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.222	78 Pt Platinum 195.084	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.383	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium 209	85 At Astatine 210	86 Rn Radon 222						
87 Fr Francium 223	88 Ra Radium 226	89-103 f- block	104 Rf Rutherfordium 261	105 Db Dubnium 262	106 Sg Seaborgium 266	107 Bh Bohrium 264	108 Hs Hassium 277	109 Mt Meitnerium 268	110 Ds Darmstadtium 271	111 Rg Roentgenium 272	112 Cn Copernicium 285	113 Nh Nihonium 284	114 Fl Flerovium 289	115 Mc Moscovium 288	116 Lv Livermorium 293	117 Ts Tennessine 289	118 Og Oganesson 294						
f- block																							
☆		57 La Lanthanum 138.905	58 Ce Cerium 140.12	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.259	69 Tm Thulium 168.930	70 Yb Ytterbium 173.054	71 Lu Lutetium 174.967							
#		89 Ac Actinium 227	90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.083	99 Es Einsteinium 252.083	100 Fm Fermium 257.105	101 Md Mendelevium 258.105	102 No Nobelium 259.108	103 Lr Lawrencium 262.109							

Periods and electronic configuration



Can you tell?

- On going through the modern periodic table it is seen that the elements Li, Be, B, C, N, O, F and Ne belong to the period-2. Write down electronic configuration of all of them.
- Is the number of valence electrons same for all these elements?
- Is the number of shells the same in these ?

You will find that the number of valence electrons is different in these elements. However, the number of shells is the same. You will also find that, while going from left to right, within the period, the atomic number increases by one at a time and the number of valence electrons also increases by one at a time.

	1	2	13	14	15	16	17	18
1	H 1							He 2
2	Li 2,1	Be 2,2	B 2,3	C 2,4	N 2,5	O 2,6	F 2,7	Ne 2,8
3	Na 2,8,1	Mg 2,8,2	Al 2,8,3	Si 2,8,4	P 2,8,5	S 2,8,6	Cl 2,8,7	Ar 2,8,8
4	K 2,8,8,1	Ca 2,8,8,2						
5		Sr						
6		Ba						
7		Ra						

Potassium atom

Argon atom

2.8 New period new shell

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Tobias Bleicker



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1. Understanding the eBook Periodic Table Physical Science If8767 Answers
 - The Rise of Digital Reading Periodic Table Physical Science If8767 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Physical Science If8767 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Physical Science If8767 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Physical Science If8767 Answers
 - Personalized Recommendations
 - Periodic Table Physical Science If8767 Answers User Reviews and Ratings
 - Periodic Table Physical Science If8767 Answers and Bestseller Lists
5. Accessing Periodic Table Physical Science If8767 Answers Free and Paid eBooks
 - Periodic Table Physical Science If8767 Answers Public Domain eBooks
 - Periodic Table Physical Science If8767 Answers eBook Subscription Services
 - Periodic Table Physical Science If8767 Answers Budget-Friendly Options

6. Navigating Periodic Table Physical Science If8767 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Physical Science If8767 Answers Compatibility with Devices
 - Periodic Table Physical Science If8767 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Physical Science If8767 Answers
 - Highlighting and Note-Taking Periodic Table Physical Science If8767 Answers
 - Interactive Elements Periodic Table Physical Science If8767 Answers
8. Staying Engaged with Periodic Table Physical Science If8767 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Physical Science If8767 Answers
9. Balancing eBooks and Physical Books Periodic Table Physical Science If8767 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Physical Science If8767 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Physical Science If8767 Answers
 - Setting Reading Goals Periodic Table Physical Science If8767 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Physical Science If8767 Answers
 - Fact-Checking eBook Content of Periodic Table Physical Science If8767 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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