

## 2.6 - Periodic Trends worksheet

Name \_\_\_\_\_

### I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

### II. Answer the questions using the symbols on the periodic table

\_\_\_\_\_ 1. What elements are in period 5?

What element(s) is a

\_\_\_\_\_ 2. halogens

\_\_\_\_\_ 3. lanthanides

\_\_\_\_\_ 4. alkaline earth metals

\_\_\_\_\_ 5. noble gases

\_\_\_\_\_ 6. transition metals

\_\_\_\_\_ 7. alkali metals

\_\_\_\_\_ 8. actinides

What elements are in the sublevel

\_\_\_\_\_ 9. s

\_\_\_\_\_ 10. p

\_\_\_\_\_ 11. d

\_\_\_\_\_ 12. f

What elements are

\_\_\_\_\_ 13. metals,

\_\_\_\_\_ 14. nonmetals

\_\_\_\_\_ 15. metalloids,

\_\_\_\_\_ 16. Which element is the most reactive nonmetal?

\_\_\_\_\_ 17. Which element is the most reactive metal?

\_\_\_\_\_ 18. Which element below is the least reactive?

\_\_\_\_\_ 19. Which element has the largest radius in period 4?

\_\_\_\_\_ 20. Which element has the largest radius in the Alkali metals family?

\_\_\_\_\_ 21. Which element has the highest ionization energy in period four?

\_\_\_\_\_ 22. Which elements has the second largest atomic number?

\_\_\_\_\_ 23. Which element has the largest atomic mass?

\_\_\_\_\_ 24. Which element is located at period 4, group 15?

\_\_\_\_\_ 25. Which element is located at period 5, group 2?

\_\_\_\_\_ 27. Which elements will form +1 ions?

\_\_\_\_\_ 28. Which group of metals will have the same properties

\_\_\_\_\_ 29. Which group of nonmetals will have the same properties

\_\_\_\_\_ 30. On the back, define all chapter 5, section 3 vocabulary terms.

b																	r		
																	h		o
	a																		u
i								t											
				e														z	

		w												
					y									



# Periodic Table Chemistry Answers

**Sasha Hu**



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## **Table of Contents Periodic Table Chemistry Answers**

1. Understanding the eBook Periodic Table Chemistry Answers
  - The Rise of Digital Reading Periodic Table Chemistry Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Chemistry Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Chemistry Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Chemistry Answers
  - Personalized Recommendations
  - Periodic Table Chemistry Answers User Reviews and Ratings
  - Periodic Table Chemistry Answers and Bestseller Lists
5. Accessing Periodic Table Chemistry Answers Free and Paid eBooks

- Periodic Table Chemistry Answers Public Domain eBooks
  - Periodic Table Chemistry Answers eBook Subscription Services
  - Periodic Table Chemistry Answers Budget-Friendly Options
6. Navigating Periodic Table Chemistry Answers eBook Formats
    - ePub, PDF, MOBI, and More
    - Periodic Table Chemistry Answers Compatibility with Devices
    - Periodic Table Chemistry Answers Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Periodic Table Chemistry Answers
    - Highlighting and Note-Taking Periodic Table Chemistry Answers
    - Interactive Elements Periodic Table Chemistry Answers
  8. Staying Engaged with Periodic Table Chemistry Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Periodic Table Chemistry Answers
  9. Balancing eBooks and Physical Books Periodic Table Chemistry Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Periodic Table Chemistry Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Periodic Table Chemistry Answers
    - Setting Reading Goals Periodic Table Chemistry Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Periodic Table Chemistry Answers
    - Fact-Checking eBook Content of Periodic Table Chemistry Answers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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