

# PERIODIC TABLE (ADVANCED)

## Molecular Structure of Hydrogen



### Basic Information

Name: Hydrogen

Symbol: H

Atomic Number: 1

Atomic Mass: 1.00794 amu

Melting Point:  $-259.14^{\circ}\text{C}$  ( $14.009185^{\circ}\text{K}$ ,  $-434.45203^{\circ}\text{F}$ )

Boiling Point:  $-252.87^{\circ}\text{C}$  ( $20.280005^{\circ}\text{K}$ ,  $-423.166^{\circ}\text{F}$ )

Number of Protons/Electrons: 1

Number of Neutrons: 0

Classification: Non-metal

Crystal Structure: Hexagonal

Density @  $293^{\circ}\text{K}$ :  $0.000089\text{ g/cm}^3$

Color: colorless

### Atomic Structure

Number of Energy Levels: 1

First Energy Level: 1

### Isotopes

Isotope	Half-life
H-1	Stable
H-2	Stable
H-3	12.3 years

### Facts

Name Origin: from words *hydro* (water) and *genes* (generated)

Uses: balloons, metal refining

Obtained From: mines, oil, gas wells

## Molecular Structure of Carbon



### Name: Carbon

Symbol: C

Atomic Number: 6

Atomic Mass: 12.0107 amu

Melting Point:  $3500.0^{\circ}\text{C}$  ( $3773.15^{\circ}\text{K}$ ,  $6332.0^{\circ}\text{F}$ )

Boiling Point:  $4827.0^{\circ}\text{C}$  ( $5100.15^{\circ}\text{K}$ ,  $8700.6^{\circ}\text{F}$ )

Number of Protons/Electrons: 6

Number of Neutrons: 6

Classification: Non-metal

Crystal Structure: Hexagonal

Density @  $293^{\circ}\text{K}$ :  $2.62\text{ g/cm}^3$

Color: May be black

### Atomic Structure

Number of Energy Levels: 2

First Energy Level: 2

Second Energy Level: 4

### Isotopes

Isotope	Half-life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.3 seconds

### Facts

Uses: steel, fibers

Obtained From: burning with insufficient oxygen

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## **Table of Contents Periodic Table Advanced Speedy Study Guide**

1. Understanding the eBook Periodic Table Advanced Speedy Study Guide
  - The Rise of Digital Reading Periodic Table Advanced Speedy Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Advanced Speedy Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Advanced Speedy Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Advanced Speedy Study Guide
  - Personalized Recommendations
  - Periodic Table Advanced Speedy Study Guide User Reviews and Ratings
  - Periodic Table Advanced Speedy Study Guide and Bestseller Lists
5. Accessing Periodic Table Advanced Speedy Study Guide Free and Paid eBooks

- Periodic Table Advanced Speedy Study Guide Public Domain eBooks
  - Periodic Table Advanced Speedy Study Guide eBook Subscription Services
  - Periodic Table Advanced Speedy Study Guide Budget-Friendly Options
6. Navigating Periodic Table Advanced Speedy Study Guide eBook Formats
    - ePub, PDF, MOBI, and More
    - Periodic Table Advanced Speedy Study Guide Compatibility with Devices
    - Periodic Table Advanced Speedy Study Guide Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Periodic Table Advanced Speedy Study Guide
    - Highlighting and Note-Taking Periodic Table Advanced Speedy Study Guide
    - Interactive Elements Periodic Table Advanced Speedy Study Guide
  8. Staying Engaged with Periodic Table Advanced Speedy Study Guide
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Periodic Table Advanced Speedy Study Guide
  9. Balancing eBooks and Physical Books Periodic Table Advanced Speedy Study Guide
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Periodic Table Advanced Speedy Study Guide
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Periodic Table Advanced Speedy Study Guide
    - Setting Reading Goals Periodic Table Advanced Speedy Study Guide
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Periodic Table Advanced Speedy Study Guide
    - Fact-Checking eBook Content of Periodic Table Advanced Speedy Study Guide
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development



- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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