

# Periodic Trends Worksheet

Name: \_\_\_\_\_ Grade: \_\_\_\_\_ ID: \_\_\_\_\_

**Directions:** Use your notes to answer the following questions.

- 1** Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.  
**Ans:** O, C, Al, K
- 2** Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.  
**Ans:** Ne, Al, S, O
- 3** Why does fluorine have a higher ionization energy than iodine?  
**Ans:** Fluorine has a smaller atomic radius, so the proton can exert a greater pull on electron (Valence shell is close)
- 4** Why do elements in the same family generally have similar properties?  
**Ans:** They have the same valence shell electron arrangement.
- 5** Indicate whether the following properties increase or decrease from left to right across the periodic table.
  - a. atomic radius (excluding noble gases) **Decrease**
  - b. first ionization energy **Increase**
  - c. electronegativity **Increase**
- 6** What trend in atomic radius occurs across the periodic table? What causes this trend?  
**Ans:** Atomic radius increases down a group because energy levels (shells) are added.
- 7** What trend in ionization energy occurs across a period on the periodic table? What causes this trend?  
**Ans:** Ionization energy increases across a period because as elements become less metallic, it requires more energy to remove an electron.
- 8** Circle the atom in each pair that has the largest radius.
  - a. Al or B
  - b. Na or Al
  - c. S or O
  - d. O or F
  - e. Br or Cl
  - f. Mg or Ca

# Periodic Table Trends Answer Key

**Mark Weller, Mark T. Weller, Tina  
Overton, Jonathan Rourke, Fraser  
Armstrong**



## **Periodic Table Trends Answer Key:**

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**Chemistry in a Month** John Wilkinson, 2003

**Shriver and Atkins' Inorganic Chemistry** Peter Atkins, 2010 Inorganic Chemistry fifth edition represents an integral part of a student s chemistry education Basic chemical principles are set out clearly in Foundations and are fully developed throughout the text culminating in the cutting edge research topics of the Frontiers which illustrate

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