				Score	/100
Mil in receiver	REV	Period	Thursday :		

Fill in the blank (2 points each blank)

- Elements that have the properties of both metal and non-metals are metalloids.
- 2. What do elements in the same group have the same number of? Valence electrons
- 3. Which element would have the highest electron affinity? Fluorine
- 4. What element is at period 3, group 8A? Argon
- 5. How many valence electrons does Bromine have? Z
- 6. The atomic number tells you how many protons (and electrons if neutral) there are in an atom.
- 7. Two Characteristics of Non-metals are not good conductors of heat or electricity, not malleable, etc.
- 8. Two characteristics of Metals are malleable, ductile, good conductors of heat and electricity
- Which element would have the smallest atomic radius? Helium
- Which element would have the highest ionization energy? Helium

11. For the questions below and based on the information provided, till in the missing information. (1 point each)

Name	Symbol	# of protons	# of electrons	# of neutrons	Atomic Number	Moss Number
Potassium	K	19	15	22	15	-41
Phosphorous	p	1.5	15	18	15	33
Beryllium	Be	4	-4	4	4	8
Sulfur	S	16	16	20	16	3.6

- 12. Draw Bohr Models for (3 points each)
 - a. Na23

b. N15

13. What is an isotope? (1 point)

An atom with a different number of neutrons

14. How can you tell the identity of any atom? (1 point)

Number of PROTONS

15. Alien Periodic Chart -- Directions: Read the clue and place the alien symbol in the corresponding box!

Periodic Table Test With Answer Key

Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan

Periodic Table Test With Answer Key:

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Thermodynamics I Bond Energies Some Commonly Used Terms in Thermodynamics The First Law of Thermodynamics Enthalpy Hess's Law of Heat Summation Standard States Heat of Vaporization and Heat of Fusion Thermodynamics II Entropy The Second Law of Thermodynamics Standard Entropies and Free Energies Electrochemistry Oxidation and Reduction Electrolytic Cells Non Standard State Cell Potentials Atomic Theory Atomic Weight Types of Bonds Periodic Trends Electronegativity Quantum Chemistry Basic Electron Charges Components of Atomic Structure The Wave Mechanical Model Subshells and Electron Configuration Double and Triple Bonds Organic Chemistry Nomenclature and Structure Alkanes Alkenes Dienes Alkynes Alkyl Halides Cyclic Hydrocarbons Aromatic Hydrocarbons Aryl Halides Ethers and Epoxides Alcohols and Glycols Carboxylic Acids Carboxylic Acid Derivatives Esters Amides Arenes Aldehydes and Ketones Amines Phenols and Quinones Structural Isomerism SIX PRACTICE EXAMS Practice Test 1 Answer Key Detailed Explanations of Answers Practice Test 2 Answer Key Detailed Explanations of Answers Practice Test 3 Answer Key Detailed Explanations of Answers Practice Test 4 Answer Key Detailed Explanations of Answers Practice Test 5 Answer Key Detailed Explanations of Answers Practice Test 6 Answer Key Detailed Explanations of Answers THE PERIODIC TABLE EXCERPT About Research OR 2 To demonstrate proficiency in Chemistry The SAT II Chemistry exam is designed for students who have taken one year of college preparatory chemistry Who Administers The Test The SAT II Chemistry Subject Test is developed by the College Board and administered by Educational Testing Service ETS The test development process involves the assistance of educators throughout the country and is designed and implemented to ensure that the content and difficulty level of the test are appropriate When Should the SAT II Chemistry be Taken If you are applying to a college that requires Subject Test scores as part of the admissions process you should take the SAT II Chemistry Subject Test toward the end of your junior year or at the beginning of your senior year If your scores are being used only for placement purposes you may be able to take the test in the spring of your senior year For more information be sure to contact the colleges to which you are applying When and Where is the Test Given The SAT II Chemistry Subject Test is administered five times a year at many locations throughout the country mostly high schools To receive information on upcoming administrations of the exam consult the publication Taking the SAT II Subject Tests which may be obtained from your guidance counselor or by contacting College Board SAT Program P O Box 6200 Princeton NJ 08541 6200 Phone 609 771 7600 Website http www collegeboard com Is There a Registration Fee Yes There is a registration fee to take the SAT II Chemistry Consult the publication Taking the SAT II Subject Tests for information on the fee structure Financial assistance may be granted in certain situations To find out if you qualify and to register for assistance contact your academic advisor HOW TO USE THIS BOOK What Do I Study First Remember that the SAT II Chemistry Subject Test is designed to test knowledge that has been acquired throughout your education Therefore the best way to prepare for the exam is to refresh yourself by thoroughly studying our review material and taking the sample tests provided in this book They will familiarize you with the types of questions directions and format

of the SAT II Chemistry Subject Test To begin your studies read over the review and the suggestions for test taking take one of the practice tests to determine your area s of weakness and then restudy the review material focusing on your specific problem areas The course review includes the information you need to know when taking the exam Be sure to take the remaining practice tests to further test yourself and become familiar with the format of the SAT II Chemistry Subject Test When Should I Start Studying It is never too early to start studying for the SAT II Chemistry test The earlier you begin the more time you will have to sharpen your skills Do not procrastinate Cramming is not an effective way to study since it does not allow you the time needed to learn the test material The sooner you learn the format of the exam the more comfortable you will be when you take the exam FORMAT OF THE SAT II CHEMISTRY The SAT II Chemistry is a one hour exam consisting of 85 multiple choice questions. The first part of the exam consists of classification questions. This question type presents a list of statements or questions that you must match up with a group of choices lettered A through E Each choice may be used once more than once or not at all The exam then shifts to relationship analysis questions which you will answer in a specially numbered section of your answer sheet You will have to determine if each of two statements is true or false and if the second statement is a correct explanation of the first The last section is composed strictly of multiple choice questions with choices lettered A through E Material Tested The following chart summarizes the distribution of topics covered on the SAT II Chemistry Subject Test Topic Percentage Number of Questions Atomic however do not review too much at any one time Concentrate on one problem area at a time by reviewing the questions and explanations and by studying our review until you are confident you completely understand the material Keep track of your scores By doing so you will be able to gauge your progress and discover general weaknesses in particular sections You should carefully study the reviews that cover your areas of difficulty as this will build your skills in those areas TEST TAKING TIPS Although you may be unfamiliar with standardized tests such as the SAT II Chemistry Subject Test there are many ways to acquaint yourself with this type of examination and help alleviate your test taking anxieties Become comfortable with the format of the exam When you are practicing to take the SAT II Chemistry Subject Test simulate the conditions under which you will be taking the actual test Stay calm and pace yourself After simulating the test only a couple of times you will boost your chances of doing well and you will be able to sit down for the actual exam with much more confidence Know the directions and format for each section of the test Familiarizing yourself with the directions and format of the exam will not only save you time but will also ensure that you are familiar enough with the SAT II Chemistry Subject Test to avoid nervousness and the mistakes caused by being nervous Do your scratchwork in the margins of the test booklet You will not be given scrap paper during the exam and you may not perform scratchwork on your answer sheet Space is provided in your test booklet to do any necessary work or draw diagrams If you are unsure of an answer guess However if you do guess guess wisely Use the process of elimination by going through each answer to a question and ruling out as many of the answer choices as possible By eliminating three answer

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Table of Contents Periodic Table Test With Answer Key

- 1. Understanding the eBook Periodic Table Test With Answer Key
 - The Rise of Digital Reading Periodic Table Test With Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Test With Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Test With Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Table Test With Answer Key
 - Personalized Recommendations

- Periodic Table Test With Answer Key User Reviews and Ratings
- Periodic Table Test With Answer Key and Bestseller Lists
- 5. Accessing Periodic Table Test With Answer Key Free and Paid eBooks
 - Periodic Table Test With Answer Key Public Domain eBooks
 - Periodic Table Test With Answer Key eBook Subscription Services
 - Periodic Table Test With Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table Test With Answer Key eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Periodic Table Test With Answer Key Compatibility with Devices
 - o Periodic Table Test With Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Test With Answer Key
 - Highlighting and Note-Taking Periodic Table Test With Answer Key
 - Interactive Elements Periodic Table Test With Answer Key
- 8. Staying Engaged with Periodic Table Test With Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Test With Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Test With Answer Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Periodic Table Test With Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Test With Answer Key
 - Setting Reading Goals Periodic Table Test With Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Test With Answer Key
 - Fact-Checking eBook Content of Periodic Table Test With Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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