

## Chemistry Exam Review Sheet

The midterm exam includes Chapters 1-9 and Chapter 21 in the Holt Modern Chemistry Book. Any additional topics that were covered in class through notes, labs, sample problems, worksheets, and etc. are also fair game for the test. The Chapter Highlights at the end of each chapter and Chapter Reviews are a great place review. Good Luck!!!

### Unit 1 – Safety, Measurement, and Classification of Matter (Chapters 1 & 2)

- **Safety**
  - See Safety Contract and Safety Lab
  - Intro pages xviii – xxi
- **Measurement with metric system and SI base units (pp 33 – 42)**
  - Practice p. 40 and p. 42
  - Section Review p. 42
  - Chapter review p. 59 # 5 – 10
- **Density w/ graphical analysis (mass vs. volume) (pp. 38 – 40)**
  - Practice p. 40
  - Chapter Review p. 59 # 16 -19
- **Significant Figures (p 46 – 50)**
  - Sample Problem D p. 47
  - Practice 1 & 2 p. 48
  - Sample Problem E p. 49
  - Practice 1 – 4 p. 50
  - Chap. Review p. 60 # 29, 30, 40, 41
- **Scientific Notation (50 – 54)**
  - Sample Problem F p. 54
  - Practice p. 54 1 – 4
  - Chapter Review p. 60 # 31, 43, 44
- **Unit Conversions w/ Dimensional Analysis (pp. 40 – 42)**
  - Be able to perform conversions between metric prefixes
- **Theory vs. Law (pp. 29 – 32)**
  - Section Review p. 31 # 1 – 5
  - Chapter Review p. 59 # 1 – 3
- **Accuracy vs. Precision (pp. 44 – 45)**
  - Practice p. 45 # 1 – 2
- **Quantitative vs. Qualitative Data (p. 29)**
  - Define and distinguish between the two, also, be able to provide examples
- **What is Chemistry? What is a chemical? (pp. 3 – 4)**
  - Section Review p. 5 # 1 – 4
  - Chapter Review p. 22 # 1 & 3
- **Extensive and Intensive Properties of Matter (p. 7)**
  - Chapter Review p. 22 # 9
- **Chemical and Physical Properties and Changes (p. 7 – 10)**
  - Section Review # 1 – 4
  - Chapter Review p. 22 # 10 – 13
- **Differentiate between Atoms, Elements, Compounds, and Mixtures**
- **Classification of Matter Flowchart**
  - Look at your notes
- **Homogenous vs. Homogeneous Mixtures (p. 12)**
  - Define and distinguish between the two, also, be able to provide examples
- **Characteristics of States of Matter and State Changes**

### Unit 2 History of the Atom, Atomic Structure w/Isotopes, Electron Configuration, Nuclear Decay, and Energy (Chapter 3 – Chapter 5 and Chapter 21)

- **History of the Atom (Democritus p. 67, Lavoisier p. 303, Mendeleev p. 133-134)**
- **Rutherford's Gold Foil Experiment (p. 74)**
- **Dalton's Atomic Theory (5 parts, some that are not correct) (p. 68)**

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**Judd E. Hollander**



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## **Table of Contents Section 2 Workbook Answers Modern Chemistry Review**

1. Understanding the eBook Section 2 Workbook Answers Modern Chemistry Review
  - The Rise of Digital Reading Section 2 Workbook Answers Modern Chemistry Review
  - Advantages of eBooks Over Traditional Books
2. Identifying Section 2 Workbook Answers Modern Chemistry Review
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Section 2 Workbook Answers Modern Chemistry Review
  - User-Friendly Interface
4. Exploring eBook Recommendations from Section 2 Workbook Answers Modern Chemistry Review

- Personalized Recommendations
  - Section 2 Workbook Answers Modern Chemistry Review User Reviews and Ratings
  - Section 2 Workbook Answers Modern Chemistry Review and Bestseller Lists
5. Accessing Section 2 Workbook Answers Modern Chemistry Review Free and Paid eBooks
    - Section 2 Workbook Answers Modern Chemistry Review Public Domain eBooks
    - Section 2 Workbook Answers Modern Chemistry Review eBook Subscription Services
    - Section 2 Workbook Answers Modern Chemistry Review Budget-Friendly Options
  6. Navigating Section 2 Workbook Answers Modern Chemistry Review eBook Formats
    - ePub, PDF, MOBI, and More
    - Section 2 Workbook Answers Modern Chemistry Review Compatibility with Devices
    - Section 2 Workbook Answers Modern Chemistry Review Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Section 2 Workbook Answers Modern Chemistry Review
    - Highlighting and Note-Taking Section 2 Workbook Answers Modern Chemistry Review
    - Interactive Elements Section 2 Workbook Answers Modern Chemistry Review
  8. Staying Engaged with Section 2 Workbook Answers Modern Chemistry Review
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Section 2 Workbook Answers Modern Chemistry Review
  9. Balancing eBooks and Physical Books Section 2 Workbook Answers Modern Chemistry Review
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Section 2 Workbook Answers Modern Chemistry Review
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Section 2 Workbook Answers Modern Chemistry Review
    - Setting Reading Goals Section 2 Workbook Answers Modern Chemistry Review
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Section 2 Workbook Answers Modern Chemistry Review



- Fact-Checking eBook Content of Section 2 Workbook Answers Modern Chemistry Review
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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