

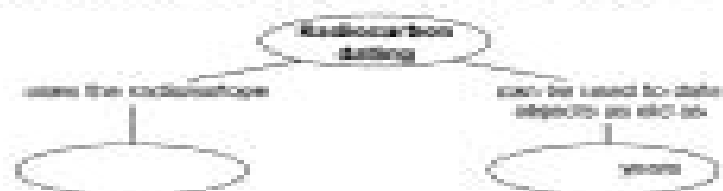
## Chapter 10 Nuclear Chemistry

**Section 10.2 Rates of Nuclear Decay****(pages 299–301)**

This section discusses half-lives and explains how nuclear decay can be used to estimate the age of objects.

**Reading Strategy (page 299)**

**Identifying Details** As you read, complete the concept map below to identify details about radiocarbon dating. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Half-life (pages 299–300)**

1. A nuclear decay rate describes \_\_\_\_\_.
2. Is the following sentence true or false? All radioisotopes decay at the same rate. \_\_\_\_\_
3. Describe a half-life. \_\_\_\_\_
4. Circle the letter that describes a sample of a radioisotope after two half-lives.
  - a. One eighth of the original sample is unchanged.
  - b. One quarter of the original sample is unchanged.
  - c. Half of the original sample is unchanged.
  - d. Three quarters of the original sample is unchanged.
5. Circle the letter of the correct answer. Iodine-131 has a half-life of 8.07 days. What fraction of a sample of iodine-131 is left unchanged after 16.14 days?
 

a. $\frac{1}{2}$	b. $\frac{1}{4}$
c. $\frac{1}{8}$	d. $\frac{1}{16}$
6. Is the following sentence true or false? Like chemical reaction rates, nuclear decay rates vary with the conditions of reaction.  
\_\_\_\_\_

## Section 2 Rates Of Nuclear Decay

**C Cleary**



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