#### Covalent Bond Practice

1. Fill in the missing information on the chart.

Element	# of Protons	# of Electrons	# of Valence Electrons	# of electrons to fill outer shell.	
Carbon	1	10	4	4	
Hydrogen	1	1	/	,	
Chlorine	17	17	7	,	
Helium	5	2	2		
Phosphorus	15	15	5	3	
Oxygen	8	8	4	2	
Sulfur	110	110	6	3	
Nitrogen	7	7	5	3	

- For each of the following covalent bonds:
  - Write the symbols for each element.
  - Draw a Lewis Dot structure for the valence shell of each element.
  - Rearrange the electrons to pair up electrons from each atom.
  - Draw circles to show the sharing of electrons between each pair of atoms.
  - Draw the bond structure using chemical symbols and lines. Use one line for each pair of electrons that is shared.
  - Write the chemical formula for each molecule.

# **Section 2 How Elements Bond Answer Key**

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