

Chapter 9 Carbon Chemistry

Section 9.2 Substituted Hydrocarbons**(pages 272–274)**

This section discusses organic compounds that contain atoms of elements other than carbon and hydrogen. It also explains the relationship between the properties of organic compounds and functional groups.

Reading Strategy (page 272)

Monitoring Your Understanding As you read, complete the table by connecting each functional group with the type of compound that contains the functional group. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Connecting Functional Groups to Types of Compounds	
Functional Group	Type of Compound
–OH	
–COOH	
–NH ₂	

- Name the two main products when methane and chlorine react.
 - _____
 - _____
- To which environmental problem have researchers connected halocarbons containing chlorine and fluorine? _____

- Describe a substituted hydrocarbon. _____

- Is the following sentence true or false? The functional group in a substituted hydrocarbon determines the properties of the compound. _____

Alcohols (page 273)

- Methanol and ethanol are two examples of a class of organic compounds called _____.
- The functional group in an alcohol is represented as –OH and is called a(n) _____ group.
- Identify two ways a halocarbon can be produced.
 - _____
 - _____

Section 2 Carbon And Hydrocarbons Answers

Chandan Sengupta



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Table of Contents Section 2 Carbon And Hydrocarbons Answers

1. Understanding the eBook Section 2 Carbon And Hydrocarbons Answers
 - The Rise of Digital Reading Section 2 Carbon And Hydrocarbons Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 2 Carbon And Hydrocarbons Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 2 Carbon And Hydrocarbons Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 2 Carbon And Hydrocarbons Answers
 - Personalized Recommendations
 - Section 2 Carbon And Hydrocarbons Answers User Reviews and Ratings
 - Section 2 Carbon And Hydrocarbons Answers and Bestseller Lists
5. Accessing Section 2 Carbon And Hydrocarbons Answers Free and Paid eBooks

- Section 2 Carbon And Hydrocarbons Answers Public Domain eBooks
 - Section 2 Carbon And Hydrocarbons Answers eBook Subscription Services
 - Section 2 Carbon And Hydrocarbons Answers Budget-Friendly Options
6. Navigating Section 2 Carbon And Hydrocarbons Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Section 2 Carbon And Hydrocarbons Answers Compatibility with Devices
 - Section 2 Carbon And Hydrocarbons Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Section 2 Carbon And Hydrocarbons Answers
 - Highlighting and Note-Taking Section 2 Carbon And Hydrocarbons Answers
 - Interactive Elements Section 2 Carbon And Hydrocarbons Answers
8. Staying Engaged with Section 2 Carbon And Hydrocarbons Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 2 Carbon And Hydrocarbons Answers
9. Balancing eBooks and Physical Books Section 2 Carbon And Hydrocarbons Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 2 Carbon And Hydrocarbons Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Section 2 Carbon And Hydrocarbons Answers
- Setting Reading Goals Section 2 Carbon And Hydrocarbons Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 2 Carbon And Hydrocarbons Answers
- Fact-Checking eBook Content of Section 2 Carbon And Hydrocarbons Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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