

## Chapter 6 Thermal Energy

### Section 1 Temperature and Heat

**A. Temperature**—related to the average kinetic energy of an object's atoms or molecules

**B. Thermal energy**—the sum of the kinetic and potential energy of all the atoms in an object

1. Thermal energy increases as temperature increases.

2. At constant temperature, thermal energy increases if mass increases.

**C. Thermal energy** that flows from something at a higher temperature to something at a lower

temperature is called **heat**.

**D. Specific heat**—amount of heat needed to raise the temperature of 1 kg of a material

by 1 degree C or K

**E. Changes in thermal energy** can be calculated as *change in thermal energy equals mass times*

*change in temperature times specific heat*.

1. When heat flows into an object and its temperature rises, the change in temperature is positive.

2. When heat flows out of an object and its temperature decreases, the change in temperature

is negative.

3. A calorimeter is used to measure specific heat.

### Section 2 Transferring Thermal Energy

**A. Conduction**—transfer of thermal energy through matter by direct contact of particles

1. Kinetic energy is transferred as particles collide.

2. Solids, particularly metals, are good heat conductors.

**B. The transfer of energy by the motion of heated particles in a fluid is called convection.**

1. Convection currents transfer heat from warmer to cooler parts of a fluid.

2. Convection currents create rain forests and deserts over different regions of Earth.

**C. Radiation**—energy transfer by electromagnetic waves

1. Some radiation is absorbed and some is reflected when it strikes a material.

2. Heat transfer by radiation is faster in a gas than in a liquid or solid.

**D. Most living things control the flow of heat by using special features such as fur, blubber, or scales.**

**E. Insulator**—material that does not let heat flow through it easily

1. Gases such as air usually make better insulators than liquids or solids.

2. A vacuum layer in a thermos is a good insulator because it contains almost no matter to

allow conduction or convection to occur.

### Section 3 Using Heat

**A. Heating systems**—warm homes and buildings

# Section 1 Temperature And Heat Answer Key

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