- 3	Directed Reading for
	Content Mastery

Section 1 = How Solutions Form Section 2 = Solubility and Concentration Section 3 = Particles in Solution

Directions: For each of the following,	write the letter of the term or	phrase that best comp	pletes the sentence.

- 1. In a solution of sugar and water, the water is the _____.
 a. solvent b. solute

 2. In the dissolving process, negative polar ends are attracted to _____ polar ends.
 a. negative b. positive

 3. Grinding a solute increases its surface area, making it dissolve more _____.
 a. slowly b. quickly

 4. A gas's solubility is best in a liquid solvent when the solution is under
 - ____ pressure.
 a. high
 b. low

 5. Different substances have ____ solubilities.
 a. different
 b. the same
 - 6. The concentration of a solution of two or more liquids is expressed as a percentage by _____.

 a. area

 b. volume
 - 7. As the temperature of a liquid solvent increases, the amount of solid solute that can be dissolved in the solvent usually _____.
 a. increases
 b. decreases
 - 8. Lines on a graph that show how much solute a solvent can hold at a given temperature are called _____.
 a. solution graph
 b. solubility curves
 - 9. In the ionization process, water pulls the molecules of a polar substance apart into _____.
 a. crystals
 b. ions
 - - 11. The more particles of a solute that are added to a solution, the _____
 the freezing point of the solution.
 a. lower
 b. higher

Sections 1, 2, and 3 (page 20)

- 1. a 7. a 8. b
- 3. b 9. l
- 3. b 4. a 10. a
- 5. a 11. a 6. b

Section 1 Reinforcement Answer Key

Cornelius Rea

Section 1 Reinforcement Answer Key:	
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• Interactive and Gamified eBooks

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