

STOICHIOMETRY

(This section explains how to calculate the number of moles required or produced, based on a stoichiometric process. It teaches you how to interpret chemical equations in terms of balancing scales, representative particles, masses, and volumes at STP.)

1. How can you determine the quantities of reactants and products in a chemical reaction?

*Source: Authors' estimates. Δ and Δ^2 denote first and second differences, respectively. Δ^2 is used to control for non-stationarity in the data.

2. Quantity usually denotes the amount of a substance expressed in grams or moles.
3. A hexagon is on the left (from 3 clockwise: 58%, 2 sides, bounds (58%, 1 top (1), 1 base (1), and 4 legs (1)). Printer is "balanced equation" has the construction of this (58%):

2000 + 2000 + 1 = 4001

4. In the following sentence, insert a subject and a predicate in the indicated positions.

- © 2004 Blackwell Publishing Ltd *Journal of Internal Medicine* 255: 103–110

10. *Explain the difference between a "strong" and a "weak" argument.*

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7. Consider the lecture of this term that tells what kind of information your CLASSMATE got from a chemical equation.

- a. stretch
- b. mass
- c. size of particles**
- d. volume
- e. number of particles

Section 12 The Arithmetic Of Equations Answers

Emilie Sanchez



Section 12 The Arithmetic Of Equations Answers:

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