

Study Guide Section 11.4 Percent Yield**Key Concepts**

- The theoretical yield of a chemical reaction is the maximum amount of product that can be produced from a given amount of reactant. Theoretical yield is calculated from the balanced chemical equation.
- The actual yield is the amount of product produced. Actual yield must be obtained through experimentation.
- Percent yield is the ratio of actual yield to theoretical yield expressed as a percent. High percent yield is important in reducing the cost of every product produced through chemical processes.

$$\text{percent yield} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$$



Section 12 4 Percent Yield Study Guide For Content Mastery

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