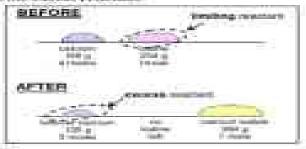
When an innercobile production plant rary out of tires no more care can be produced even if there are still planty of engages, busines, seath, and other parts. When one of the reactions in a chemical solution rary out the reaction stops as well. The substance that hunting production in a chemical reaction is known as the limiting reaction. The reaction that is left over when the reaction stops is called the excess trustment.

Consider the electrical reaction between calcium and testion. When 160 g (4 spelies) of calcium marks with 254 g (1 mole) of useling the maction makes 294 g (1 mole) of calcium todals. Abbreagh there is energy calcium to make more uniciam testion, the author runs out that Since all of the author gets used up it is called the limiting reaction. According to the testing equation, if our mole of isolate spaces, one mole of calcium will much. This means that there are still 3 moles of calcium left. Because calcium is left over it is called the success tractum.



Example

partitions a publishment for each market

. Instanton the contestions

F Configuration of the Paris Strategies - 18

whichever execute makes how product to the liveries resource.

How is the Senting reserver. There is exceeds nutting to shake 17.22 g. FeS. But units recently then by make 15.24 g of Pas.

Find the amount of product formed in each of the following reactions.

- What man of H₂O will be produced if 9.5 g of H₂ to are with 1.2 g of O₂?
 H₂ = O₂ H₂O
- If 1.85 g of MgCOth, reacts with 3.71 g of HCL. flow much MgCi, is produced? What is the limiting reaction?
 MgCOth + BCI - MgCU + BLO
 - What mans of AgCl is produced whom ET-42 g of AgNO, much with 14.79 g of NaCl? AgNO, a NaCl -- AgCl + NaNo.
 - If 14.7 g of national is placed in 11.5 g of water, what mass of hydrogen gas is produced?
 Ca + Hot → CarOtto + H.
 - What ream of princeium hydroxide is formed when 8.2 g of petacoum coude is added to 1.3 g of water?
 K.O = H.O → KOH

- What mean of alternative objective could be made from it.l g of attretterm and +2 L of obligate at 5 PP?
 Al + Cl₂ -+ AbCL
- B 5.26 L of nitrogen monoroide and 7.64 L of oxygen are conditional, what train of introgen dioxide in formed? Assume conditions are STP. NO + O₁ -> SO₂
- B. D. Lig of subscent retractionals remain with 8.4 L. of hydrogen at STP, what mass of allies in formed? SiGL + 8L -- Bis 1902.
- What map of CarOSS, a produced in the reaction of 2.6 g CarOO₂, with 6.2 g of KCHS?
 CarOO₂ + KOHE → CarOSS₁ + KNO₂
- What more of earliers disnade is formed when 64 kg of oreigner?
 C.H. + Ch. CO. + BGO

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