Balancing Chemical Equations Worksheet

1.
$$2 H_2 + 0_2 \rightarrow 2 H_2O$$

3.
$$S_8 + 12 O_2 \rightarrow 8 SO_3$$

4.
$$2 N_2 + 0_2 \rightarrow 2 N_20$$

5.
$$2 \text{ HgO} \rightarrow 2 \text{ Hg} + 0$$

6.
$$6 \, CO_2 + 6 \, H_2O \Rightarrow C_6H_{12}O_6 + 6 \, O_2$$

7.
$$Zn + Z HCI \rightarrow ZnCl_2 + H_2$$

9.
$$\underline{2}$$
 Na + $\underline{2}$ H₂O \Rightarrow $\underline{2}$ NaOH + $\underline{}$ H₂

12. ____
$$CO_2$$
 + ___ Z $NH_3 \rightarrow ____ $OC(NH_2)_2$ + ____ $H_2O$$

13.
$$_{-}^{4}$$
 Si₂H₃ + $_{-}^{7}$ O₂ \rightarrow $_{-}^{8}$ SiO₂ + $_{-}^{6}$ H₂O₃

14.
$$2 \text{ Al}(OH)_3 + 3 \text{ H}_2SO_4 \rightarrow Al_2(SO_4)_3 + 6 \text{ H}_2O$$

17.
$$2 C_7H_6O_2 + 15 O_2 \rightarrow 14 CO_2 + 5 H_2O$$

19. _____ FeS₂ + _____ O₂
$$\Rightarrow$$
 _____ Fe₂O₃ + _____ SO₂

21. ____ Fe₂O₃ + __3_ H₂
$$\rightarrow$$
 __2_ Fe + __3_ H₂O

22. ____ Na₂CO₃ + __2_ HCl
$$\rightarrow$$
 2 NaCl + ____ H₂O + ___ CO₂

24. ____C₇H₁₆ + _//_O₂
$$\rightarrow$$
 _7 CO₂ + _8 H₂O

25.
$$P_4 + 5 O_2 \rightarrow 2 P_2O_5$$

Section 10 1 Reactions And Equations Answer Key

Brian H. Gilding, Robert Kersner

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