

17

THERMOCHEMISTRY

SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510)

This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.

► Energy Transformations (page 505)

1. What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? _____
2. Where the use of energy is concerned (in a scientific sense), when is work done?

3. Circle the letter next to each sentence that is true about energy.
 - a. Energy is the capacity for doing work or supplying heat.
 - b. Energy is detected only because of its effects.
 - c. Heat is energy that transfers from one object to another because they are at the same temperature.
 - d. Gasoline contains a significant amount of chemical potential energy.
4. Circle the letter next to each sentence that is true about heat.
 - a. One effect of adding heat to a substance is an increase in the temperature of that substance.
 - b. Heat always flows from a cooler object to a warmer object.
 - c. If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperature of both objects is the same.

► Exothermic and Endothermic Processes (pages 506–507)

5. What can be considered the “system” and what are the “surroundings” when studying a mixture of chemicals undergoing a reaction? Write your answers where indicated below.

System: _____

Surroundings: _____

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RD Boyd



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