

MLA Research Paper (Daly)

Daly 1

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Professor Chavez
English 101
14 March XXXX

A Call to Action:

Regulate Use of Cell Phones on the Road

When a cell phone goes off in a classroom or at a concert, we are irritated, but at least our lives are not endangered. When we are on the road, however, irresponsible cell phone users are more than irritating: They are putting our lives at risk. Many of us have witnessed drivers so distracted by dialing and chatting that they resemble drunk drivers, weaving between lanes, for example, or nearly running down pedestrians in crosswalks. A number of bills to regulate use of cell phones on the road have been introduced in state legislatures, and the time has come to push for their passage. Regulation is needed because drivers using phones are seriously impaired and because laws on negligent and reckless driving are not sufficient to punish offenders.

No one can deny that cell phones have caused traffic deaths and injuries. Cell phones were implicated in three fatal accidents in November 1999 alone. Early in November, two-year-old Morgan Pena was killed by a driver distracted by his cell phone. Morgan's mother, Patti Pena, reports that the driver "ran a stop sign at 45 mph, broadsided my vehicle and killed Morgan as she sat in her car seat." A week later, corrections officer Shannon Smith, who was guarding prisoners by the side of the road, was killed by a woman distracted by a phone call (Besthoff). On Thanksgiving weekend that same month, John and Carole Hall were killed when a Naval

Title is centered.

Opening sentences catch readers' attention.

Thesis asserts Angela Daly's main point.

Daly uses a clear topic sentence.

Signal phrase names the author of the quotation to follow.

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Marginal annotations indicate **MLA-style formatting** and **effective writing**.

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the

titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1
 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
 one mole of HCl would be fully neutralized by one mole of $NaOH$

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being $NaOH$ reaction with a 2nd substance $H_2C_2O_4(aq) + 2NaOH(aq) \rightarrow Na_2C_2O_4 + 4H_2O$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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