

PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H Hydrogen 1.008	2 He Helium 4.003																
3 Li Lithium 6.941	4 Be Beryllium 9.012																
5 B Boron 10.81	6 C Carbon 12.01	7 N Nitrogen 14.01	8 O Oxygen 16.00	9 F Fluorine 18.99	10 Ne Neon 20.18											17 Cl Chlorine 35.45	18 Ar Argon 39.95
11 Na Sodium 22.99	12 Mg Magnesium 24.31															19 K Potassium 39.10	20 Ca Calcium 40.08
21 Sc Scandium 44.96	22 Ti Titanium 47.88	23 V Vanadium 50.94	24 Cr Chromium 52.00	25 Mn Manganese 54.94	26 Fe Iron 55.85	27 Co Cobalt 58.93	28 Ni Nickel 58.69	29 Cu Copper 63.55	30 Zn Zinc 65.38								
39 Y Yttrium 88.91	40 Zr Zirconium 91.22	41 Nb Niobium 92.91	42 Mo Molybdenum 95.94	43 Tc Technetium 98.91	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.87	48 Cd Cadmium 112.41	49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.60	53 I Iodine 126.91	54 Xe Xenon 131.29		
55 Cs Cesium 132.91	56 Ba Barium 137.33															85 At Astatine 210	86 Rn Radon 222
87 Fr Francium 223	88 Ra Radium 226															113 Nh Nihonium 286	114 Fl Flerovium 289

Atomic #	Symbol	State	Boiling Point
1	H	Solid	-252.87°C
2	He	Liquid	-268.9°C
3	Li	Solid	1342°C
4	Be	Solid	2970°C
5	B	Solid	2550°C
6	C	Solid	3500°C
7	N	Gas	-195.8°C
8	O	Gas	-183°C
9	F	Gas	-188°C
10	Ne	Gas	-269°C
11	Na	Solid	893°C
12	Mg	Solid	1363°C
13	Al	Solid	2470°C
14	Si	Solid	2355°C
15	P	Solid	280°C
16	S	Solid	444.6°C
17	Cl	Gas	-34.6°C
18	Ar	Gas	-182.3°C
19	K	Solid	770°C
20	Ca	Solid	1484°C
21	Sc	Solid	2835°C
22	Ti	Solid	3260°C
23	V	Solid	3407°C
24	Cr	Solid	2671°C
25	Mn	Solid	2071°C
26	Fe	Solid	2750°C
27	Co	Solid	2709°C
28	Ni	Solid	2730°C
29	Cu	Solid	2562°C
30	Zn	Solid	2484°C
31	Ga	Solid	24.04°C
32	Ge	Solid	937.4°C
33	As	Solid	611°C
34	Se	Solid	685.4°C
35	Br	Liquid	-7.2°C
36	Kr	Gas	-153.3°C
37	Rb	Solid	39.3°C
38	Sr	Solid	1377°C
39	Y	Solid	2700°C
40	Zr	Solid	3550°C
41	Nb	Solid	2477°C
42	Mo	Solid	2622°C
43	Tc	Solid	2916°C
44	Ru	Solid	2700°C
45	Rh	Solid	2687°C
46	Pd	Solid	2535°C
47	Ag	Solid	2162°C
48	Cd	Solid	321°C
49	In	Solid	156.6°C
50	Sn	Solid	227°C
51	Sb	Solid	477.7°C
52	Te	Solid	717°C
53	I	Solid	184.3°C
54	Xe	Gas	-108.1°C
55	Cs	Solid	28.48°C
56	Ba	Solid	1770°C
57	La	Solid	912°C
58	Ce	Solid	1068°C
59	Pr	Solid	935°C
60	Nd	Solid	912°C
61	Pm	Solid	913°C
62	Sm	Solid	774°C
63	Eu	Solid	822°C
64	Gd	Solid	1312°C
65	Tb	Solid	1354°C
66	Dy	Solid	1352°C
67	Ho	Solid	1347°C
68	Er	Solid	1328°C
69	Tm	Solid	1329°C
70	Yb	Solid	1330°C
71	Lu	Solid	1323°C
72	Hf	Solid	2536°C
73	Ta	Solid	2996°C
74	W	Solid	3422°C
75	Re	Solid	3186°C
76	Os	Solid	3054°C
77	Ir	Solid	2709°C
78	Pt	Solid	2542°C
79	Au	Solid	2856°C
80	Hg	Liquid	356.73°C
81	Tl	Solid	304°C
82	Pb	Solid	327.3°C
83	Bi	Solid	271.5°C
84	Po	Solid	253.7°C
85	At	Solid	269.7°C
86	Rn	Gas	-210°C
87	Fr	Solid	253°C
88	Ra	Solid	2010°C
89	Ac	Solid	1500°C
90	Th	Solid	1780°C
91	Pa	Solid	2512°C
92	U	Solid	2850°C
93	Np	Solid	2900°C
94	Pu	Solid	3230°C
95	Am	Solid	2880°C
96	Cm	Solid	2700°C
97	Bk	Solid	2700°C
98	Cf	Solid	2350°C
99	Es	Solid	2300°C
100	Fm	Solid	2200°C
101	Mn	Solid	2100°C
102	Lr	Solid	2000°C
103	Uuo	Solid	2000°C

Nonmetals

Other nonmetals

Metals

Alkali metals

Alkaline earth metals

Lanthanoids

Transition metals

Post-transition metals

Halogens

Noble gases

Atomic #

Symbol

State

Boiling Point

1

H

Solid

-252.87°C

2

He

Liquid

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912°C

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913°C

62

Sm

Solid

774°C

63

Eu

Solid

822°C

64

Gd

Solid

1312°C

65

Tb

Solid

1354°C

66

Dy

Solid

1352°C

67

Ho

Solid

1347°C

68

Er

Solid

1328°C

69

Tm

Solid

1329°C

70

Yb

Solid

1330°C

71

Lu

Solid

1323°C

72

Hf

Solid

2536°C

73

Ta

Solid

2996°C

74

W

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3422°C

75

Re

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Pa

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2512°C

92

U

Solid

2850°C

93

Np

Solid

2900°C

94

Pu

Solid

3230°C

95

Am

Solid

2880°C

96

Cm

Solid

2700°C

97

Bk

Solid

2700°C

98

Cf

Solid

2350°C

99

Es

Solid

2300°C

100

Fm

Solid

2200°C

101

Mn

Solid

2100°C

102

Lr

Solid

2000°C

103

Uuo

Solid

2000°C

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic form (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol 1H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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Christian G. Meyer



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