
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, she placed that question's number statement in her answer's box to question.

1. In developing his periodic table, Mendeleev's observations were
 - a. accurate.
 - b. incorrect and significant.
 - c. lacking.
 - d. **incomplete.**
2. Mendeleev's periodic table showed all elements in order of
 - a. increasing atomic mass.
 - b. increasing positive charge.
 - c. **decreasing atomic mass.**
 - d. increasing.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable elements.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on increasing
 - a. atomic number.
 - b. **atomic mass.**
 - c. atomic weight.
 - d. electronegativity.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. molecules.
 - c. ions.
 - d. compounds.

Periodic Law Section Quiz Answers

Mocktime Publication



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Unveiling the Magic of Words: A Overview of "**Periodic Law Section Quiz Answers**"

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