

Density Worksheet

Physical Science

$D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

- The ratio of an object's mass to its _____ is called the *density* of the object.
- A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
- For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - mass = 453 g; volume = 225 cm³
 - mass = 5.0 g; volume = 10.0 cm³
 - mass = 26.1 g; volume = 2.0 mL
- If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
- A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Section 2 Pollution Meeting Individual Needs Answer Key

JS Bruner



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