

Half Life Worksheet #2

- 1) Fluorine-21 has a half life of approximately 5 seconds. What fraction of the original nuclei would remain after 1 minute?

$$60 \text{ seconds} / 5 \text{ seconds} = 12 \text{ half lives}$$

$$1 \rightarrow \frac{1}{2} \rightarrow \frac{1}{4} \rightarrow \frac{1}{8} \rightarrow \frac{1}{16} \rightarrow \frac{1}{32} \rightarrow \frac{1}{64} \rightarrow \frac{1}{128} \rightarrow \frac{1}{256} \rightarrow \frac{1}{512} \rightarrow \frac{1}{1024} \rightarrow \frac{1}{2048} \rightarrow \frac{1}{4096} \text{ will remain}$$

- 2) Iodine-131 has a half life of 8 days. What fraction of the original sample would remain at the end of 32 days?

$$32 \text{ days} / 8 \text{ days} = 4 \text{ half lives}$$

$$1 \rightarrow \frac{1}{2} \rightarrow \frac{1}{4} \rightarrow \frac{1}{8} \rightarrow \frac{1}{16} \text{ will remain}$$

- 3) The half-life of chromium-51 is 28 days. If the sample contained 510 grams, how much chromium would remain after 56 days? How much would remain after 1 year? How much was present 168 days ago?

$$56/28 = 2 \text{ half lives}$$

$$510 \rightarrow 255 \rightarrow \frac{127.5}{1} \text{ grams}$$

$$365/28 = 13 \text{ half lives} \quad 510 \rightarrow 255 \rightarrow 127.5 \rightarrow 63.75 \rightarrow 31.875 \rightarrow 15.9375 \rightarrow 7.96875 \rightarrow 3.9843 \rightarrow 1.9922 \rightarrow 0.996 \rightarrow 0.498 \rightarrow 0.249 \rightarrow 0.124 \rightarrow 0.062 \text{ grams}$$

$$168/28 = 6 \text{ half lives}$$

$$510 \rightarrow 255 \rightarrow 127.5 \rightarrow 63.75 \rightarrow 31.875 \rightarrow 15.9375 \rightarrow \frac{7.97}{1} \text{ grams}$$

- 4) If 20.0 g of a radioactive isotope are present at 1:00 PM and 5.0 g remain at 2:00 PM, what is the half life of the isotope?

$$20 \rightarrow 10 \rightarrow 5 \text{ minutes}$$

$$2 \text{ half lives}$$

$$1\text{pm} - 2\text{pm} = 1 \text{ hour} / 2 = 30.$$

- 5) The half life of Uranium-238 is 4.5 billion years and the age of earth is 4.5×10^9 years. What fraction of Uranium-238 that was present when Earth was formed still remains?

$$4.5 \text{ billion} / 4.5 \text{ billion} = 1 \text{ half life has happened}$$

$$\frac{1}{2} \text{ still remains}$$

- 6) Chromium-48 decays. After 6 half-lives, what fraction of the original nuclei would remain?

$$1 \rightarrow \frac{1}{2} \rightarrow \frac{1}{4} \rightarrow \frac{1}{8} \rightarrow \frac{1}{16} \rightarrow \frac{1}{32} \rightarrow \frac{1}{64}$$

- 7) The half life of iodine-125 is 60 days. What fraction of iodine-125 nuclides would be left after 360 days?

Section 25 2 Half Lofe Worksheet

Michael Seilmaier

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03

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