

Name:	
Grade & Section:	
Subject:	

## Nuclear Decay

### Types of Radiation

1. Complete the table below to compare the properties of alpha, beta, positron, and gamma radiation.

Name	Alpha	Beta	Positron	Gamma
Greek Letter				
Symbol				
Charge				
Stopped by				

### Nuclear Decay Reactions

In problem 2-5, fill in the missing particle for each nuclear decay reaction. Then, match each equation to the appropriate type of nuclear decay.

- A. alpha emission
- B. beta emission
- C. electron capture
- D. positron emission



In problem 6 and 7, write a balanced equation for each for each nuclear decay reaction.

6. Decay of Radium-223 by alpha emission



7. Decay of Iodine-126 by positron emission



8. How are the atomic number and mass number of a nuclide affected by positron emission?
9. How are the atomic number and mass number of a nuclide affected by the emission of gamma radiation?
10. Which has a slightly greater mass in a nuclear reaction, the reactants or the product? Why?

## Section 25 2 Concept Review Nuclear Decay

**N Noddings**



## **Section 25.2 Concept Review Nuclear Decay:**

## Reviewing **Section 25 2 Concept Review Nuclear Decay**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Section 25 2 Concept Review Nuclear Decay**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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