



## Reinforcement

# Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula of hydrogen peroxide?

It is represented as the decomposition of the breaking of molecules of this chemical substance, the products are molecules of  $\text{H}_2\text{O}$  and the molecules of  $\text{O}_2$  gas.

2. What is the chemical formula of hydrogen peroxide? It is represented as  $\text{H}_2\text{O}_2$ . It is represented as  $\text{H}_2\text{O}_2$ .

3. What is the chemical formula of hydrogen peroxide and the products?

Hydrogen peroxide is a chemical compound with the molecular formula  $\text{H}_2\text{O}_2$ . It is a colorless liquid with a strong odor. It is a powerful oxidizing agent and is used in many chemical reactions. It is also used in the production of hydrogen peroxide.

4. According to the law of conservation of mass, if the mass of the reactants is 100 grams, what is the mass of the products?

100 grams



5. What is the chemical formula of the product?  $\text{H}_2\text{O}$

6. What is the chemical formula of the reactant?  $\text{H}_2\text{O}_2$

7. What is the chemical formula of the product?  $\text{H}_2\text{O}$

Hydrogen peroxide is a chemical compound with the molecular formula  $\text{H}_2\text{O}_2$ . It is a colorless liquid with a strong odor. It is a powerful oxidizing agent and is used in many chemical reactions. It is also used in the production of hydrogen peroxide.

8. What is the chemical formula of the product?  $\text{H}_2\text{O}$

9. What is the chemical formula of the product?  $\text{H}_2\text{O}$

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10. According to the law of conservation of mass, if the mass of the reactants is 100 grams, what is the mass of the products?

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**JA Banks**

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