LAB 9: Polarization

Results (Polarization)

The polarimeter is used for the determination of the concentration of optically active substances. For this a sodium light with a wavelength of 589 (nm) is linearly polarized by means of a polarizer. This light passes through the solution to be investigated and is observed through an analyzer. Optically active substances rotate the polarization plane. By measuring the rotational angle by means of the analyzer, the concentration of the solution can now be calculated using the following relation:

$$s = \frac{\theta}{LC}$$
. 0×5

- a) Determination of the length of the tube(optical path)
 L = 200 (mm) = 2 (dm)
- b) Optical zeroing (calibration)
- 1. Switch on the sodium lamp and wait a short time for it to warm up.
- 2. Turn the analyzer with the adjustment screw to 0° so that the circle is uniformly lit (dark), i.e. so that no central line is visible. When the analyzer is turned about the zero position the central line and the two edge areas become light or dark. For this reason deviations from the zero position can very easily be observed.
- c) Filling a round cuvette (tube)
- Hold the tube vertical so that the thicker part is at the top.
- 2. Take off the upper screw ring; remove the internal cap, sealing ring and glass disc.
- 3. Fill the tube with the liquid to be investigated with as few bubbles as possible.
- 4. Replace the glass disc, sealing ring and internal cap and tighten with the screw ring.
- d) Determination of the angle of rotation (θ)
- Place the tube with the liquid to be investigated into the sample chamber Turn the analyzer with the adjustment screw in such a way that the observed circle is again uniformly (dark).
- 2. Read the angle of rotation θ from the scale by means of the vernier scale

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Table of Contents Polarization Answers Sheet

- 1. Understanding the eBook Polarization Answers Sheet
 - The Rise of Digital Reading Polarization Answers Sheet
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Polarization Answers Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Polarization Answers Sheet
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Polarization Answers Sheet
 - Personalized Recommendations
 - $\circ\,$ Polarization Answers Sheet User Reviews and Ratings
 - Polarization Answers Sheet and Bestseller Lists
- 5. Accessing Polarization Answers Sheet Free and Paid eBooks
 - Polarization Answers Sheet Public Domain eBooks
 - Polarization Answers Sheet eBook Subscription Services
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 - Highlighting and Note-Taking Polarization Answers Sheet
 - Interactive Elements Polarization Answers Sheet
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polarization Answers Sheet
- 9. Balancing eBooks and Physical Books Polarization Answers Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polarization Answers Sheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Polarization Answers Sheet
 - Setting Reading Goals Polarization Answers Sheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Polarization Answers Sheet
 - Fact-Checking eBook Content of Polarization Answers Sheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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